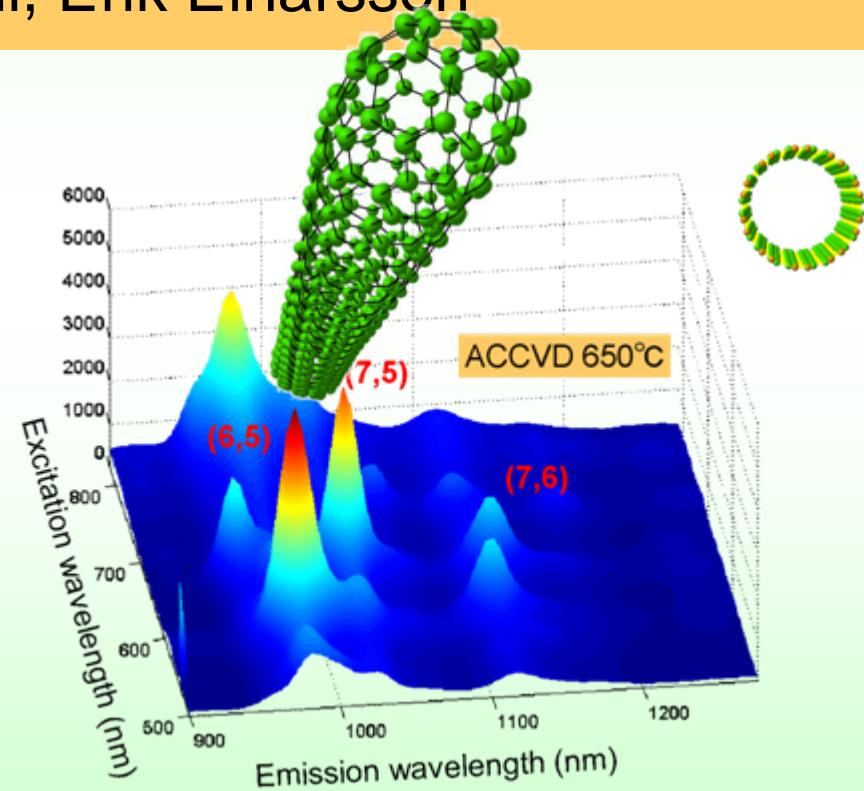
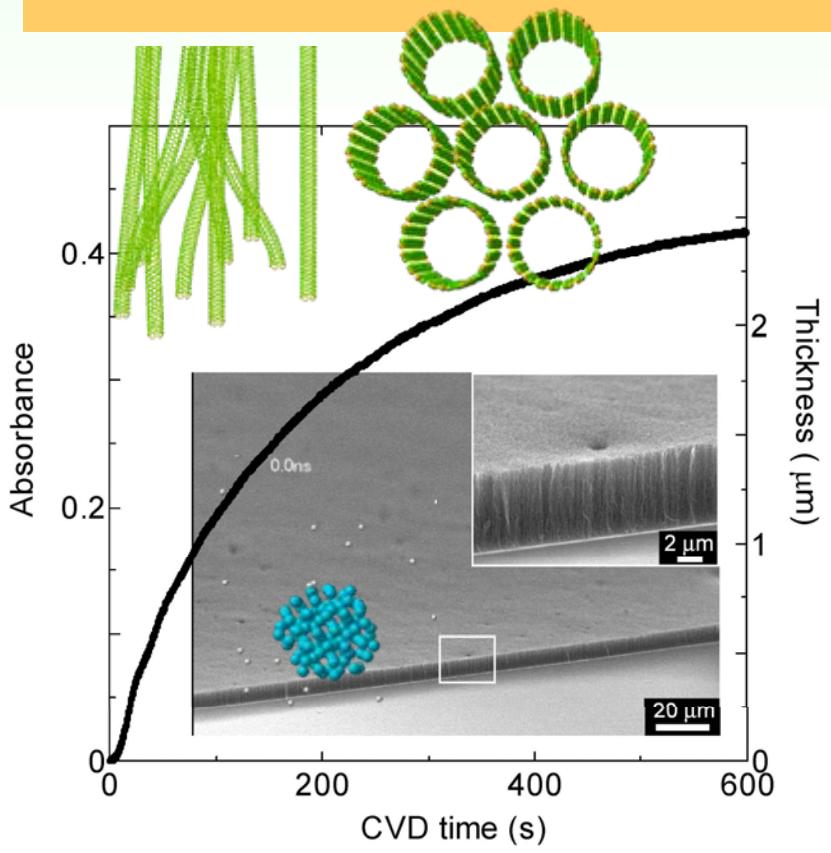


CVD Growth Mechanism of Single-Walled Carbon Nanotubes from Alcohol

Shigeo Maruyama, Yuhei Miyauchi, Yoichi Murakami,
Shohei Chiashi, Erik Einarsson



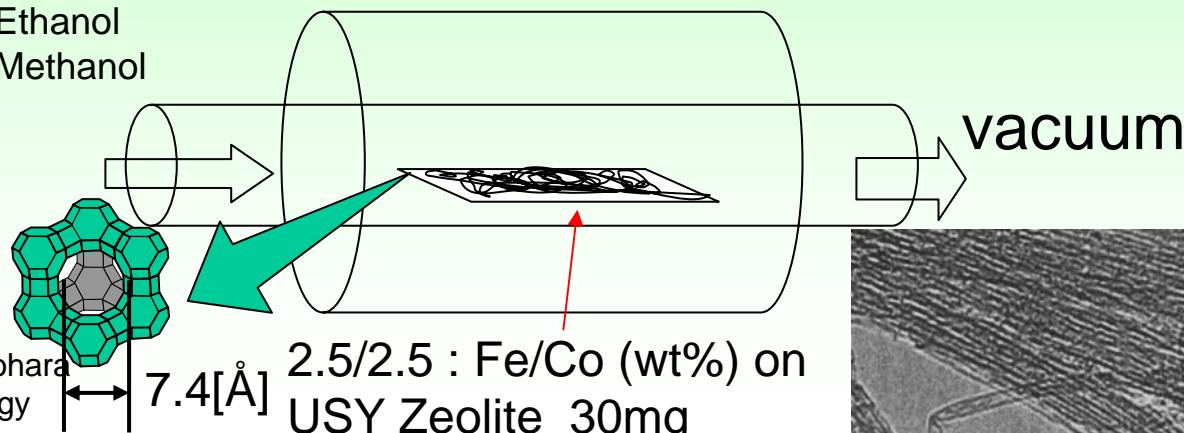
Department of Mechanical Engineering
The University of Tokyo

Alcohol CCVD on Catalysts Supported with Zeolite

Alcohol

Ethanol
Methanol

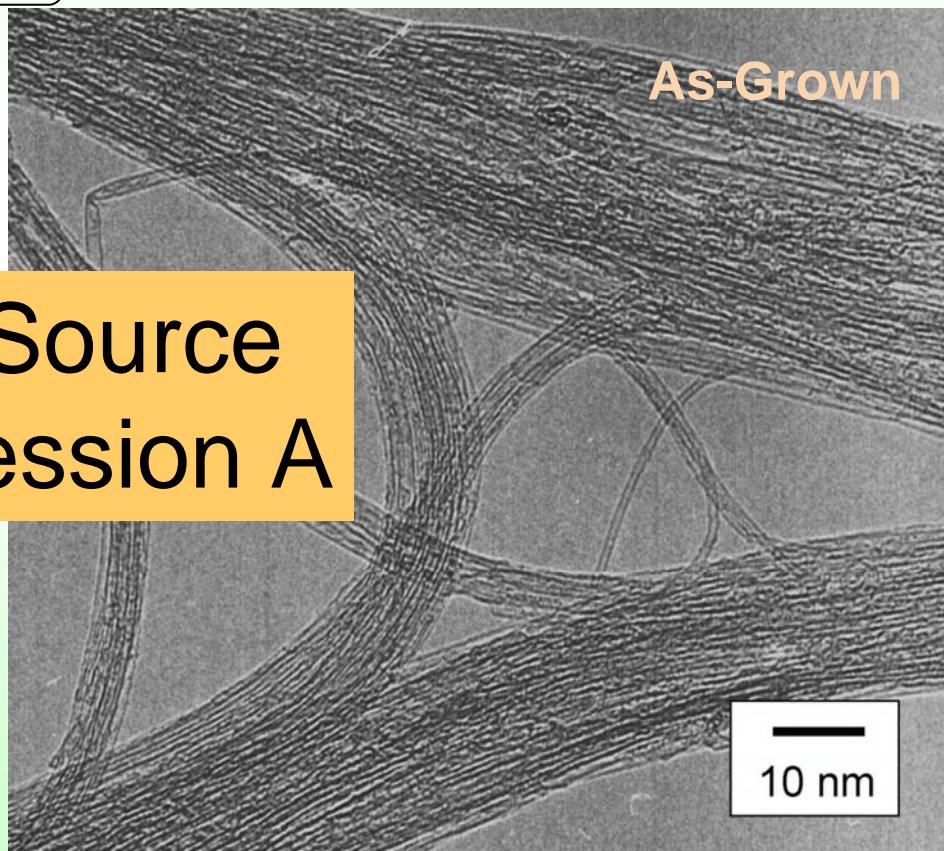
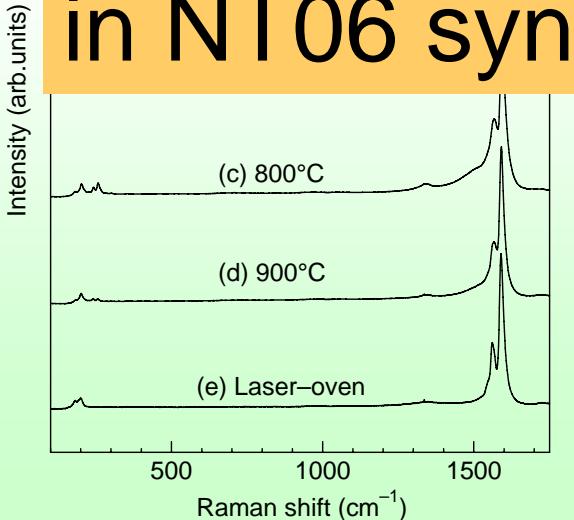
Electric Furnace



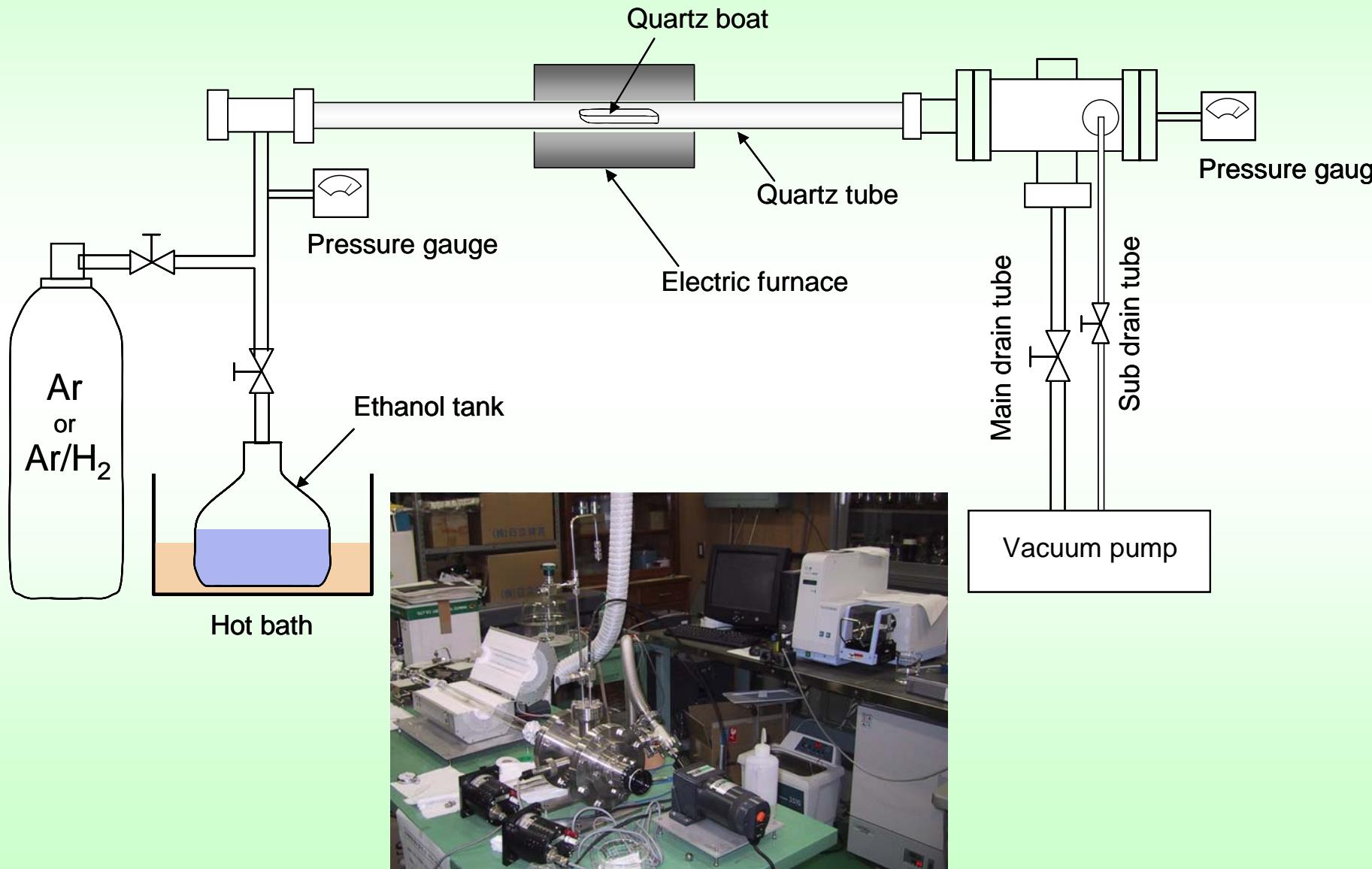
Simple, High-Purity
Low-Temperature
(550-900°C)

As-Grown

Most Used Carbon Source
in NT06 synthesis session A

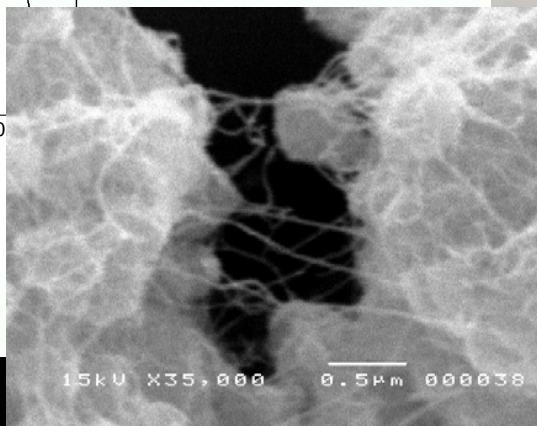
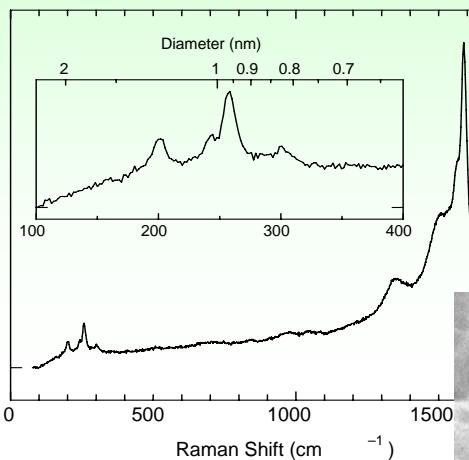


ACCVD Experimental Apparatus



ACCVD is Simple!!

High School Students Presented
at Fullerene-Nanotubes General Symposium @ Univ. of Tokyo, July 28, 2004



Synthesis of Single-walled Carbon Nanotubes from Alcohol by CCVD Method

○Miki Kanao, Kieko Sawaguchi and Masaru Moriyasu
東京工業大学工学部附属工業高等学校

try Course, Technical High School
(the School of Engineering,
Institute of Technology)

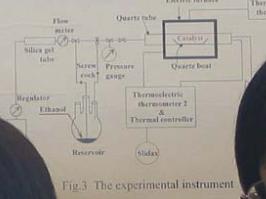
The preparation of the catalyst system and specimen
set-up in this study were based on the work by
Maruyama et al.

Formation of the Catalyst

The ethanol solution was sonicated and heated at 80 °C, and dried.
Fig.1 The ethanol solution of the catalyst

was ->
Fig.2 The catalyst

The Experimental Instrument



Results and Discussion

- Catalyst
 - Reactants
 - Thermal conditions
- The blackened reactants on the boat
Difference of colors depends on its division of the boat
A B
D E
The 6 divisions of the boat

SEM Image of the Samples

ring-like cobweb seem to be SWNTs

The SEM Image

The SEM image of the samples



Miki Kanao
High School
Tokyo Tech
July 28, 2004

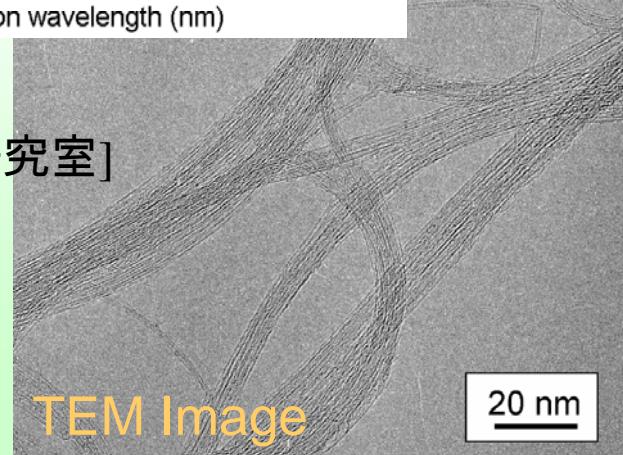
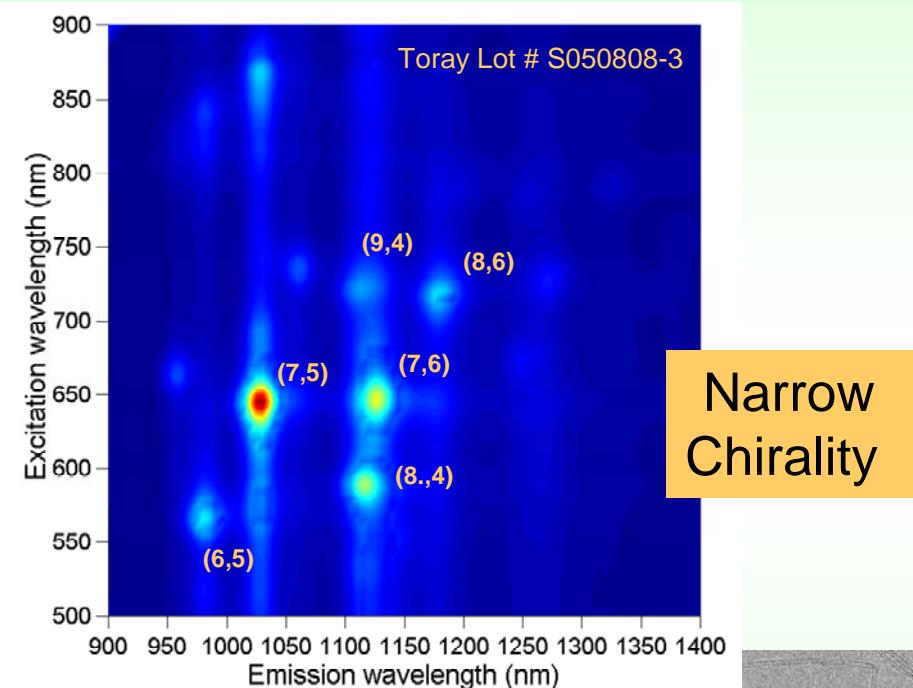


Large Scale ACCVD Production by Toray

High purity SWNTs from Ethanol on Zeolite



Purified 10 g sample



Toray Industries, Inc. [東レ株式会社]
Chemicals Research Laboratories [化成品研究所ケミカル研究室]

9-1, Oe-cho, Minato-ku, Nagoya 455-8502 JAPAN

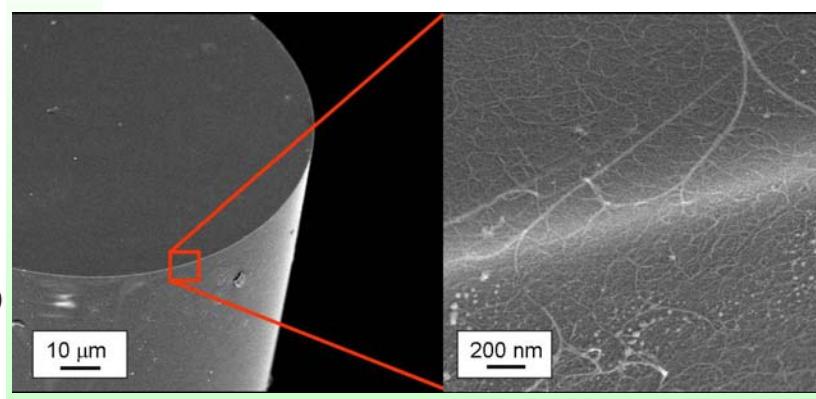
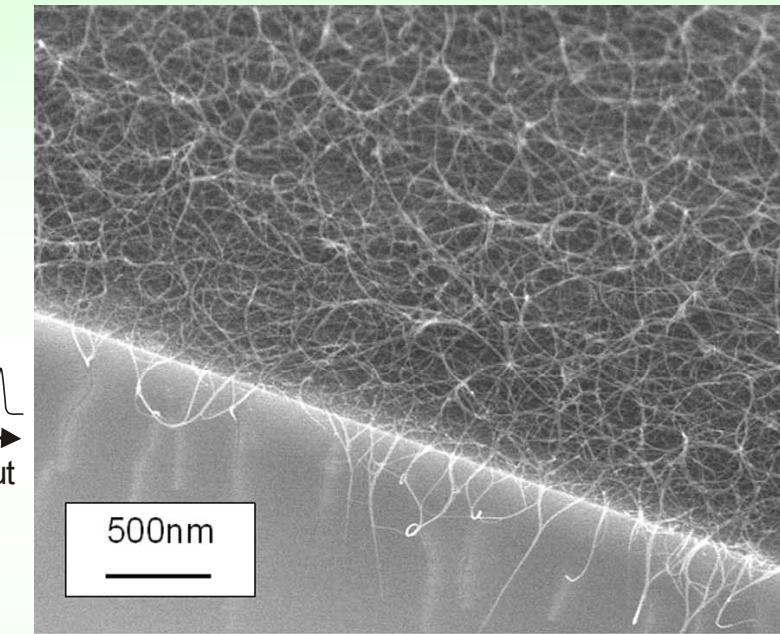
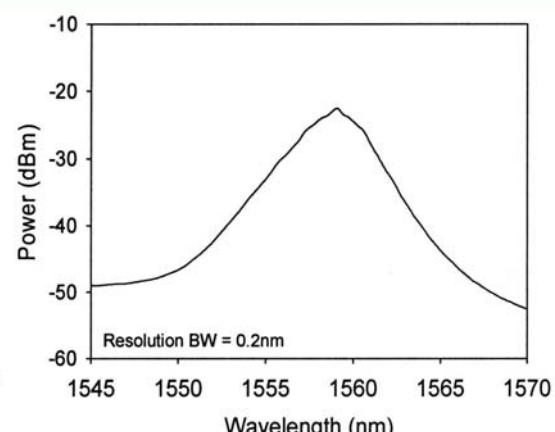
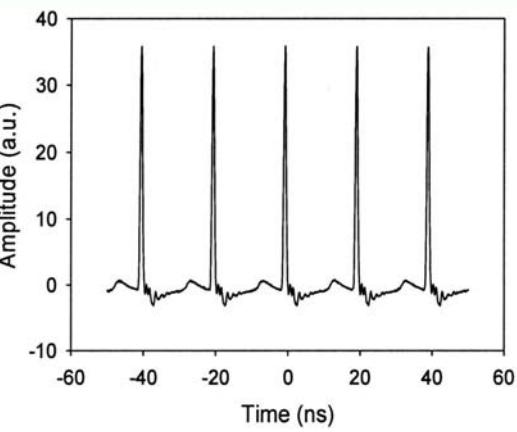
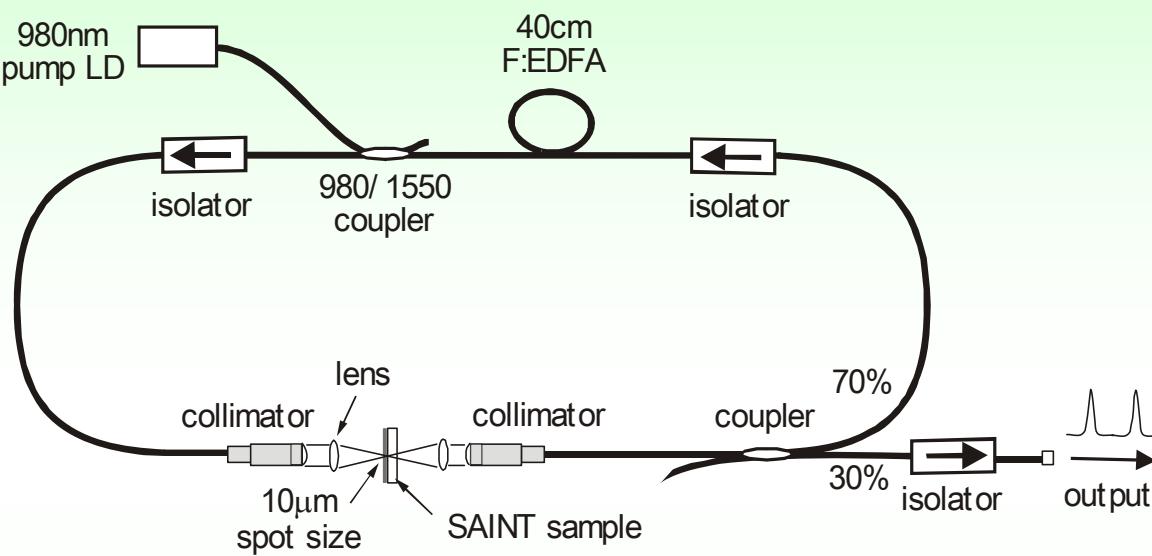
URL:<http://www.toray.co.jp>

Contact: Yuji Ozeki (Senior Research Chemist)[尾関雄治]

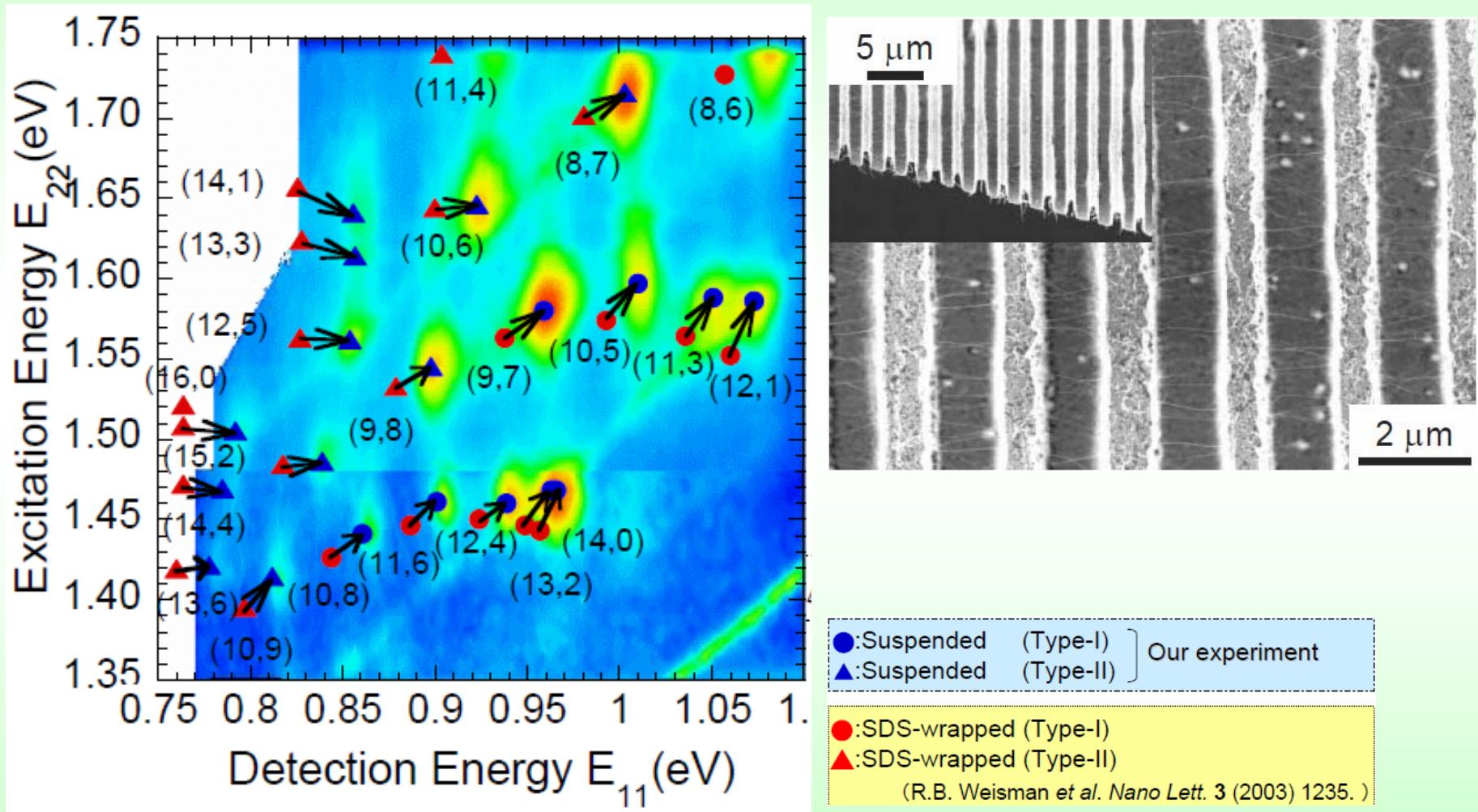
TEL +81 (52)-613-5276, FAX +81 (52)-613-5347

e-mail: yuji_ozeki@nts.toray.co.jp

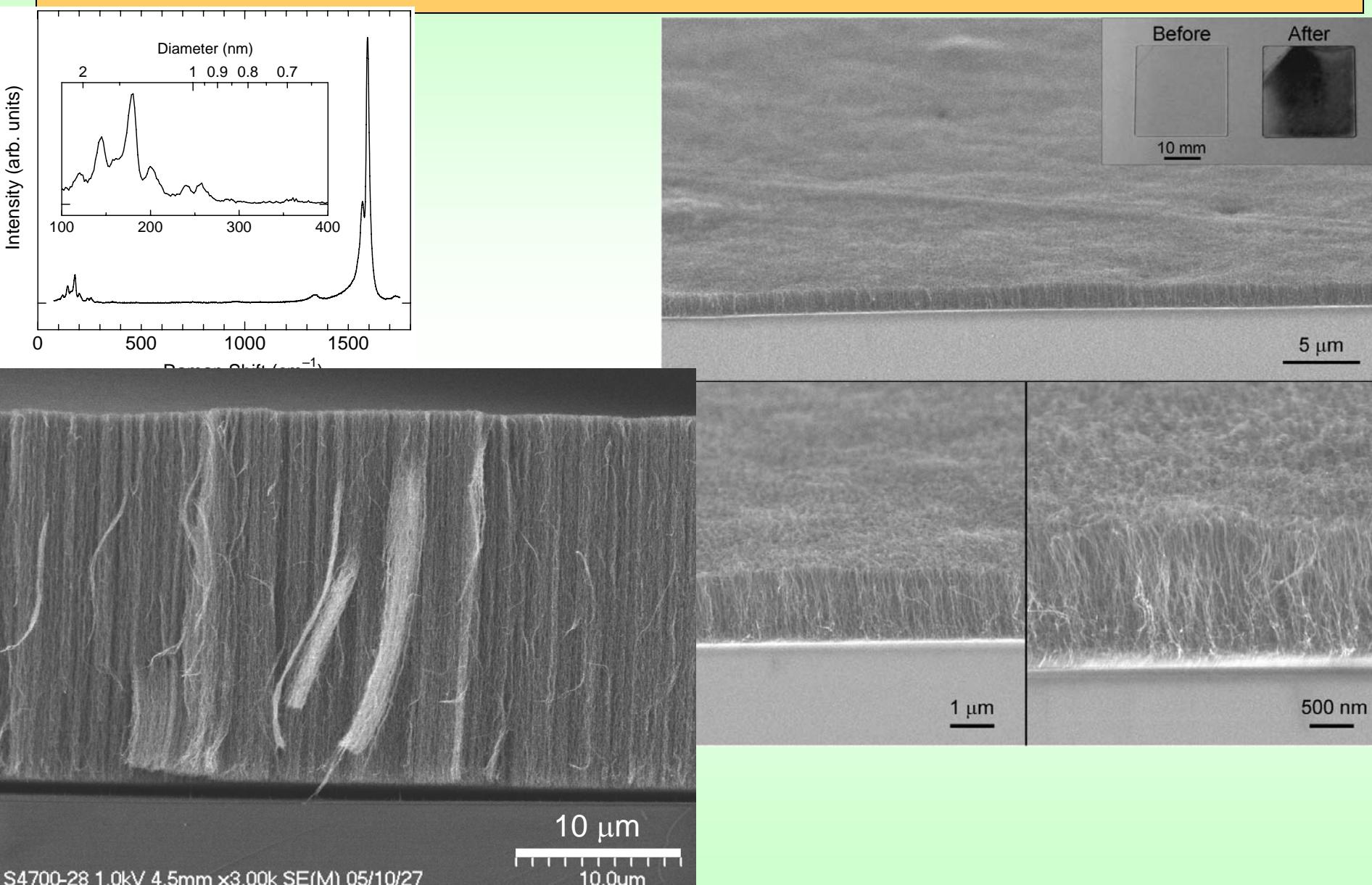
Saturable Absorbers: Application to Mode-Locked Fiber Lasers



Photoluminescence of SWNTs suspended in air



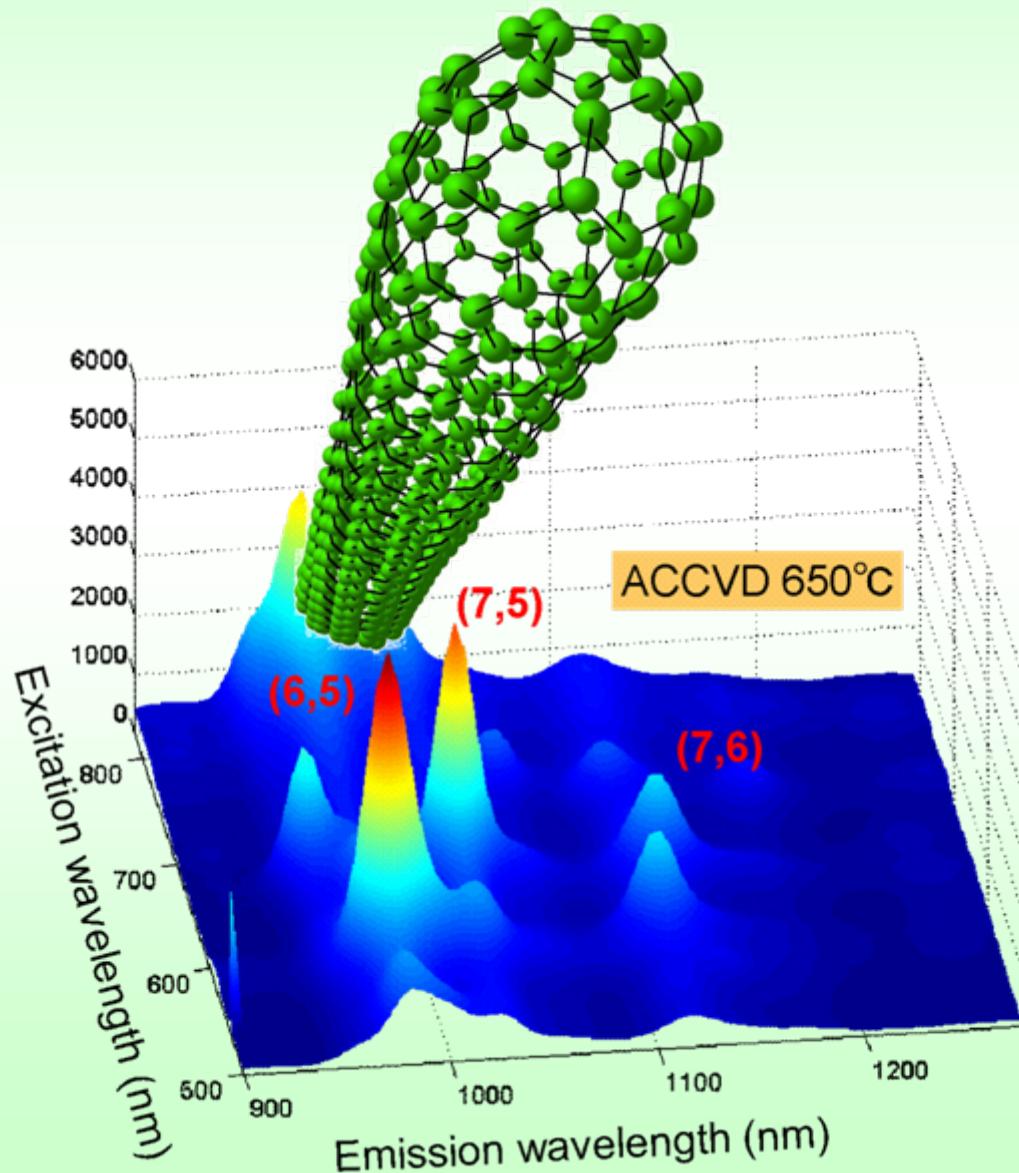
Vertically Aligned SWNTs on Quartz Substrate



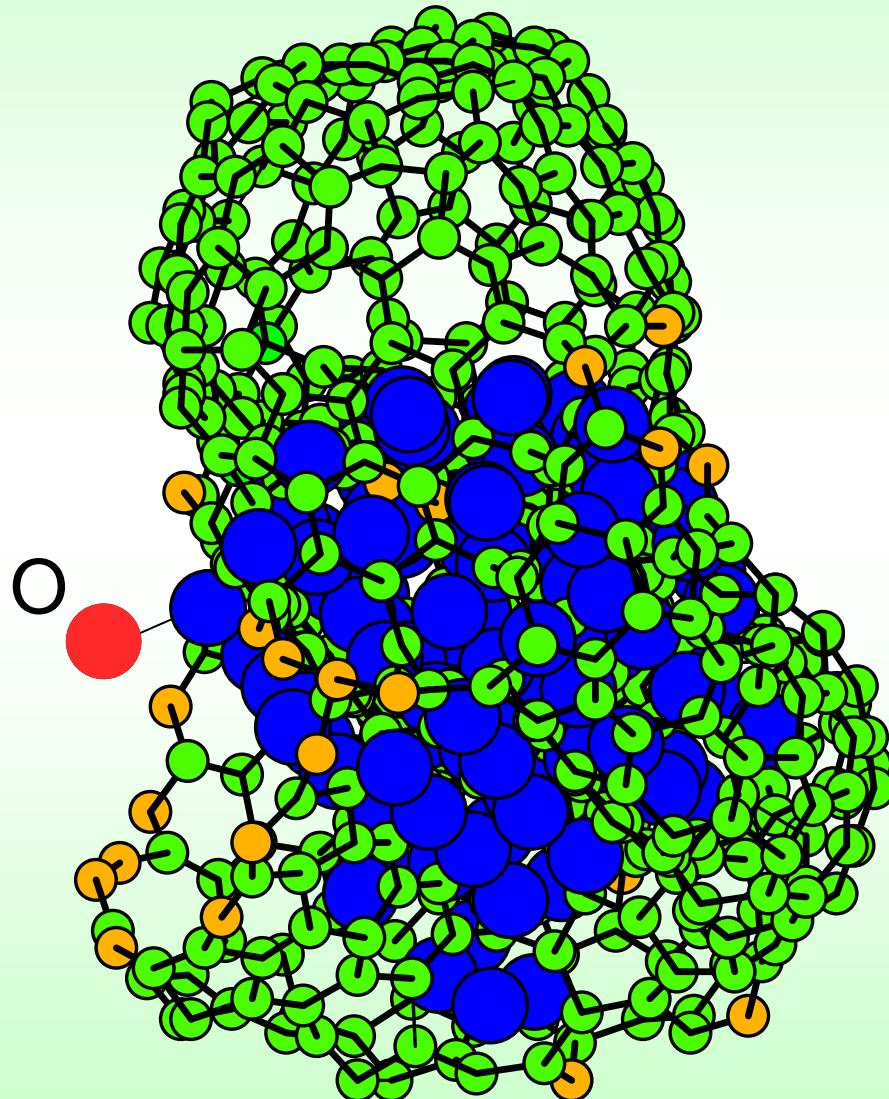
S4700-28 1.0kV 4.5mm x3.00k SE(M) 05/10/27

Y. Murakami, S. Chiashi, Y. Miyauchi, M. Hu, M. Ogura, T. Okubo, S. Maruyama, Chem. Phys. Lett. 385 (2004) 298

Narrow Chirality Growth

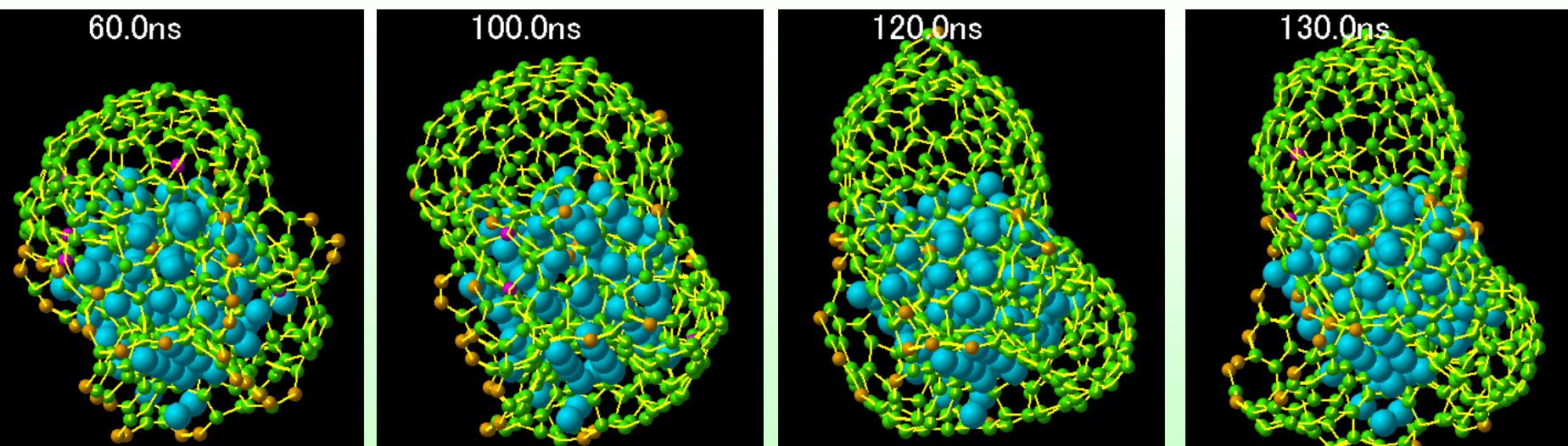
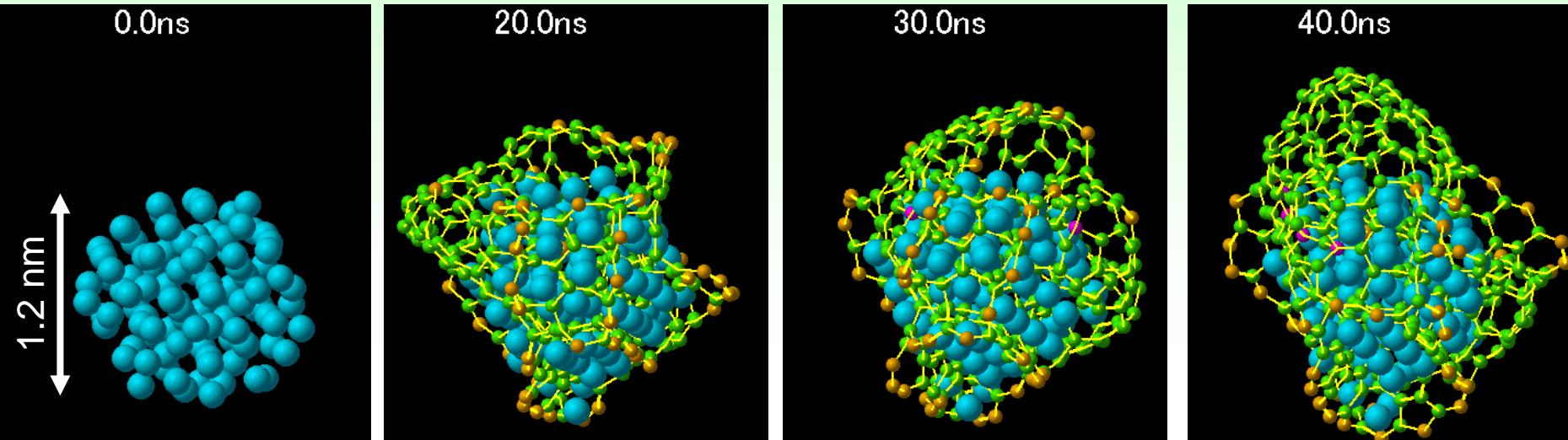


Catalysts and Alcohol Catalytic CVD Method

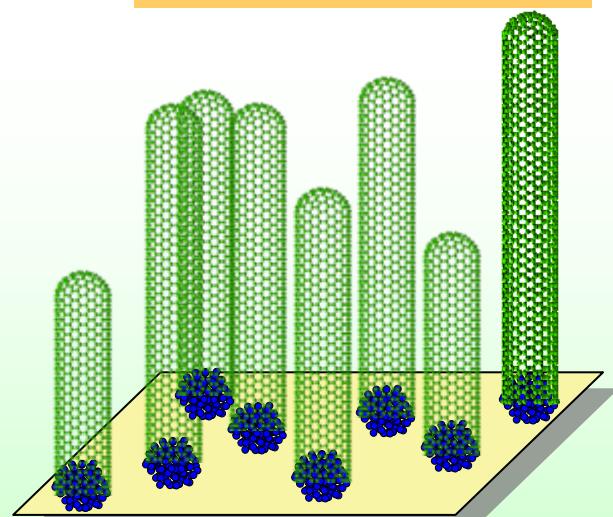
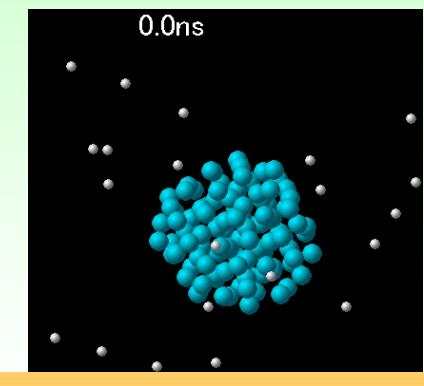
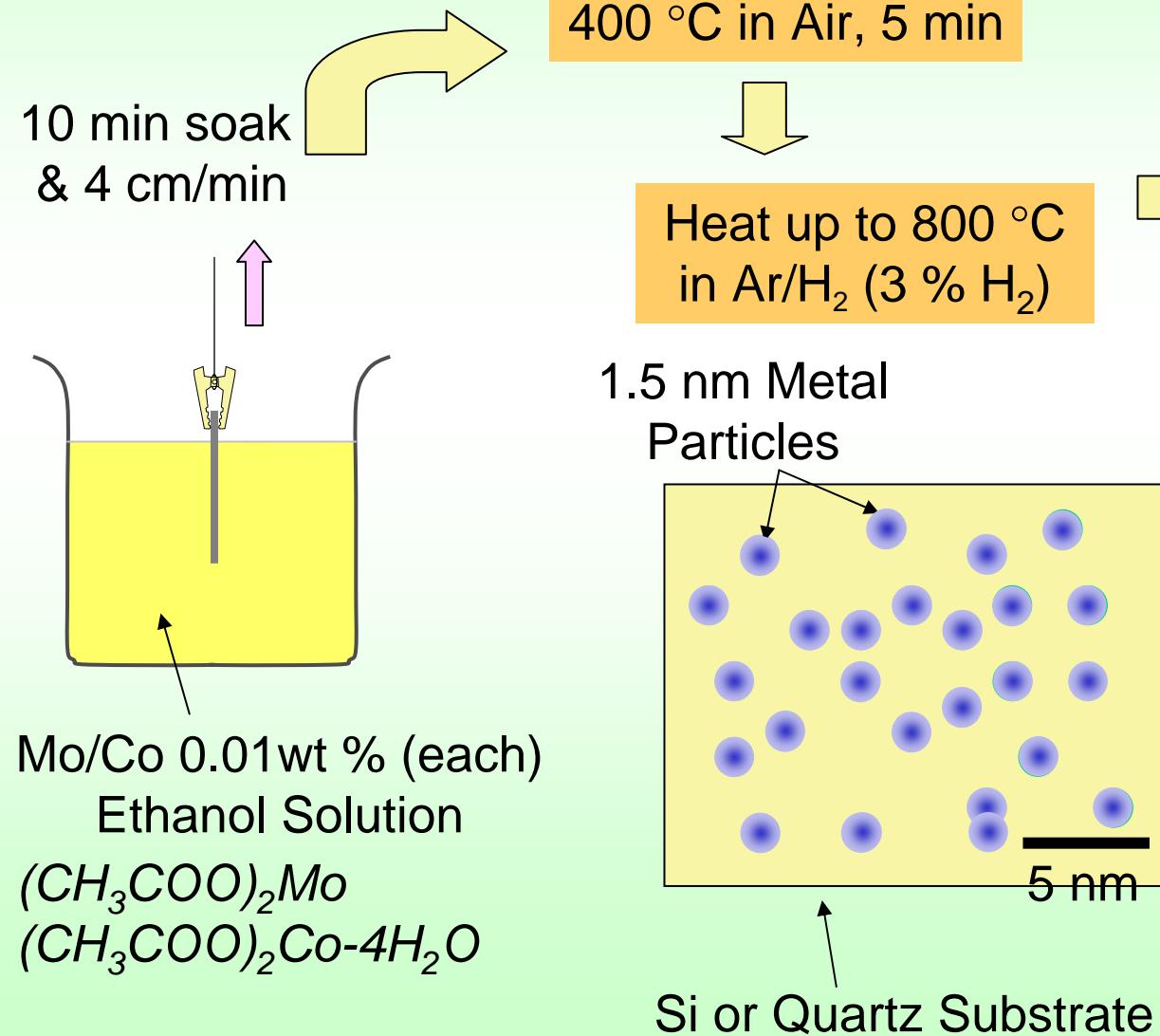


Molecular Dynamics Simulation for Mechanism

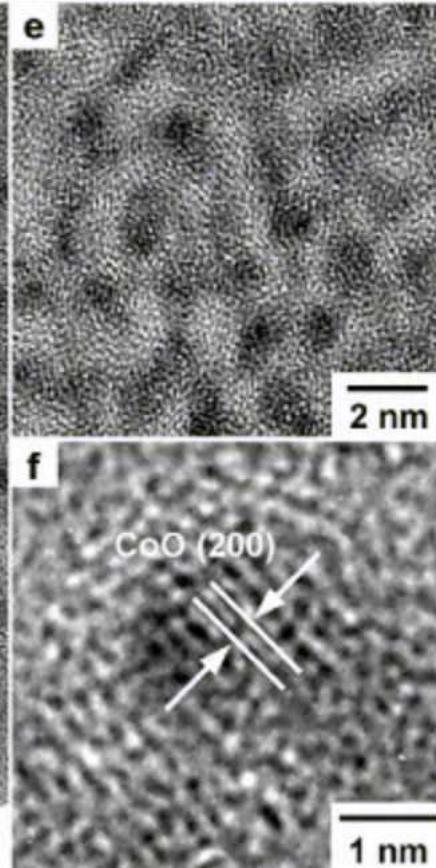
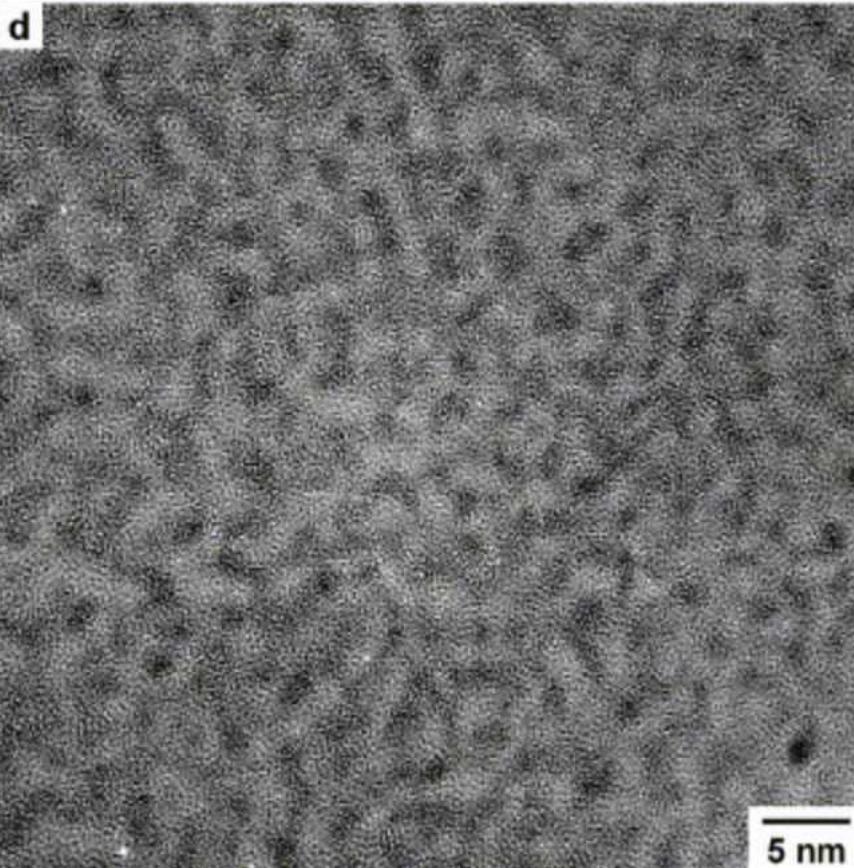
500 Carbon & Ni₁₀₈ : 2500K



ACCVD Directly on Flat Surfaces (Dip-Coat)

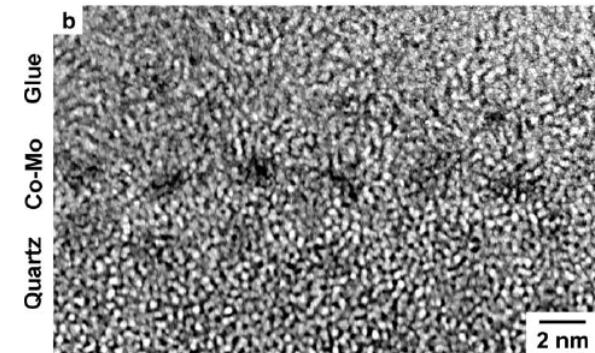


TEM images of Co-Mo catalysts on SiO₂/Si after reduction

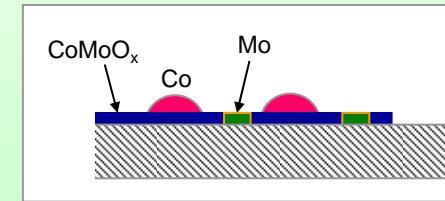


Plan TEM [Quartz/catalyst/SiO₂(10 nm)]

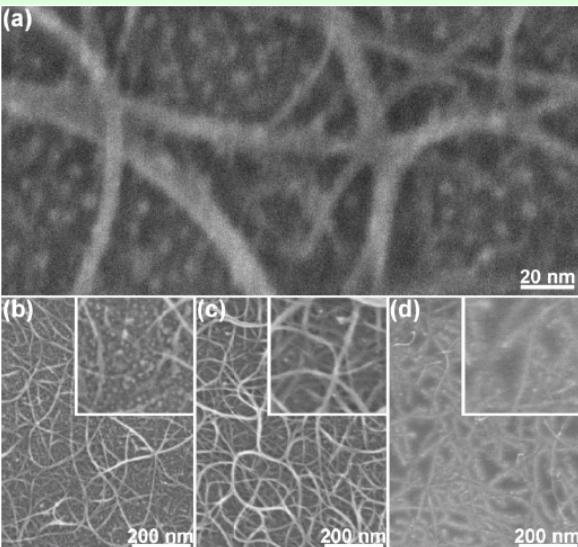
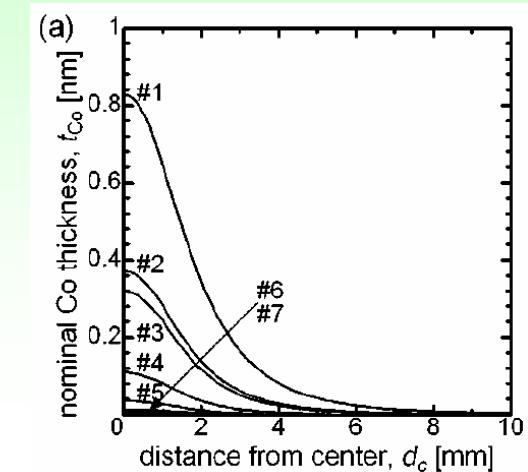
Slicing & Punching
Mech. Grinding
Ar Ion Milling



XTEM
[Quartz/catalyst/glue
/catalyst/Quartz]

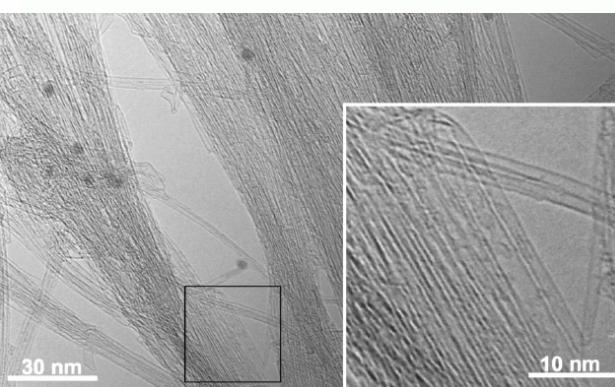
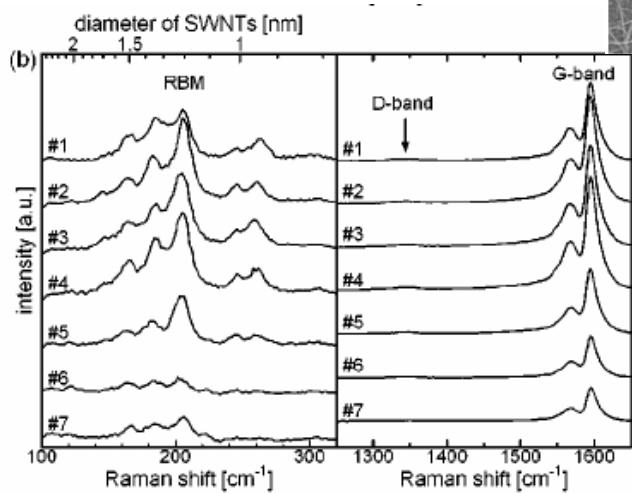


Combinatorial Method to Prepare Catalysts Ni Grows SWNTs



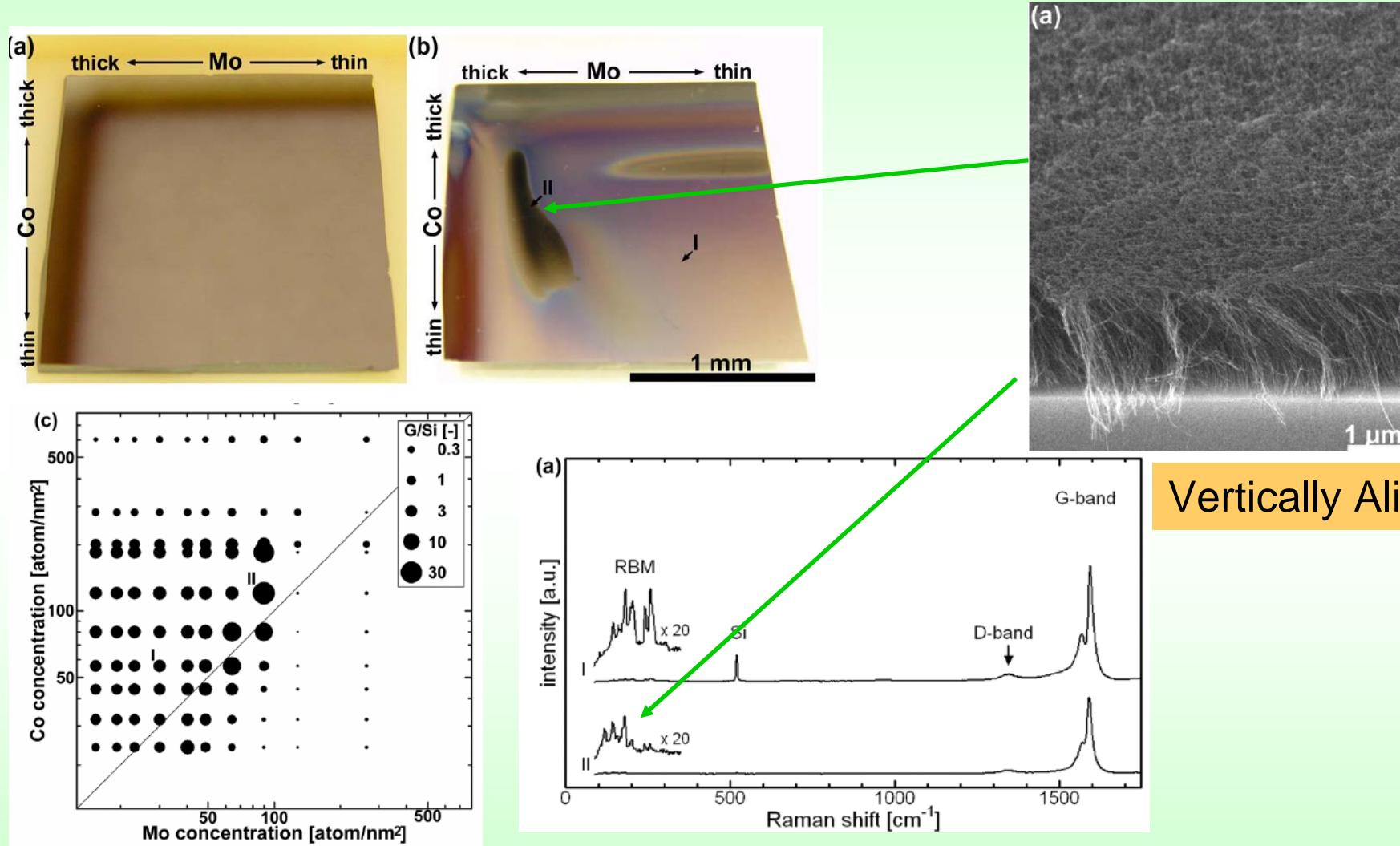
Best Condition

Co: $t_{Co}=0.081$ nm



Ni: $t_{Ni} = 0.22$ nm

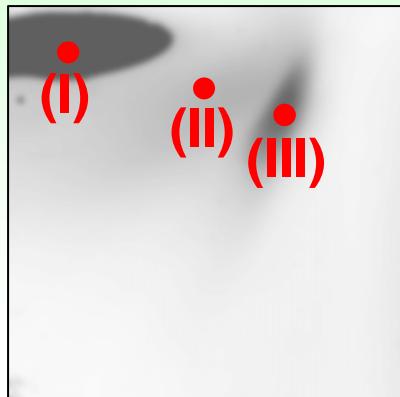
Combinatorial Method for (Co-Mo) Catalysts



S. Noda, H. Sugime, T. Osawa, Y. Tsuji, S. Chiashi, Y. Murakami, and S. Maruyama, Carbon 44 (2006) 1414.

Combinatorial Method for (Co-Mo) Catalysts (2)

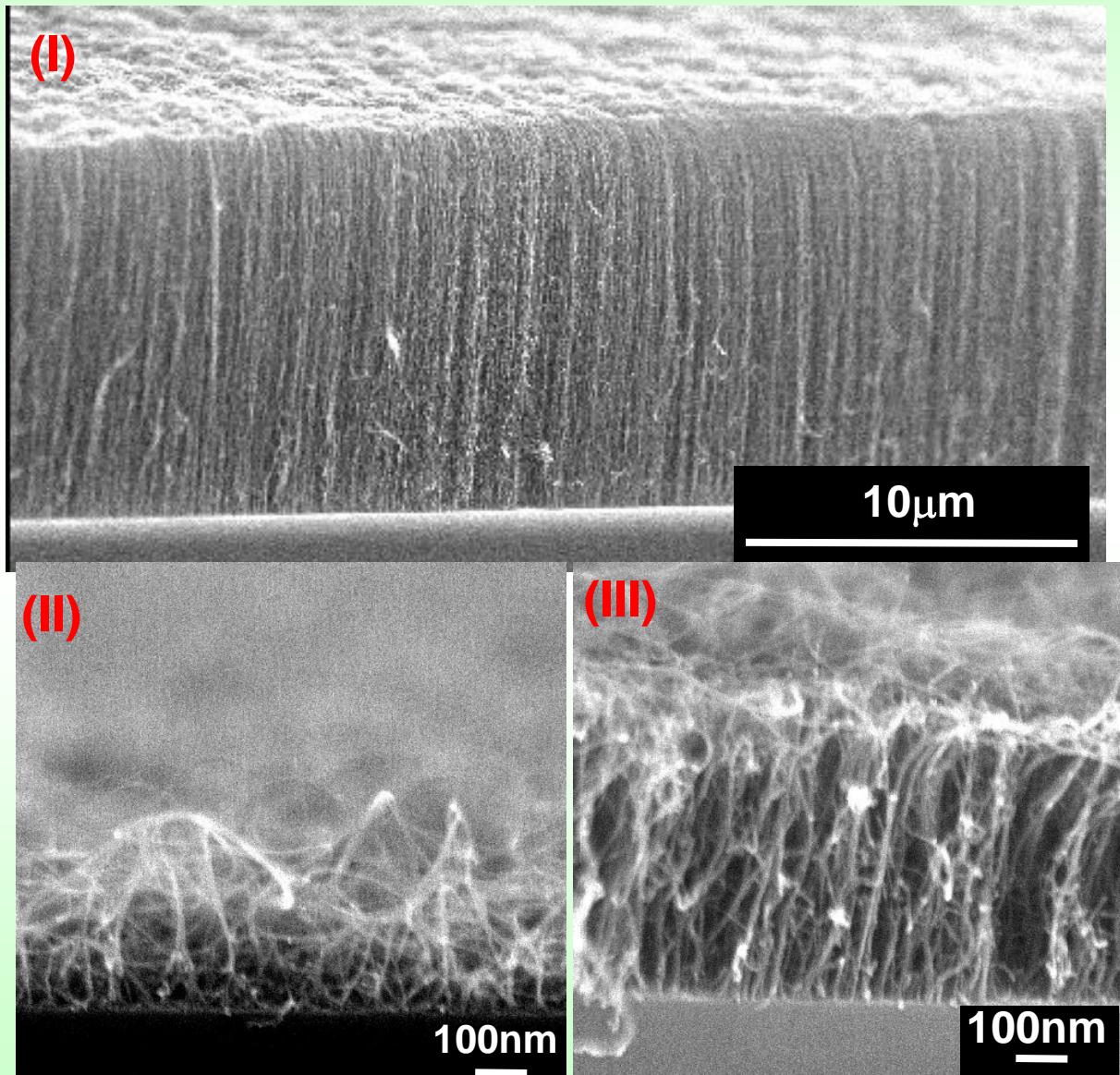
thick
↑
Co
↓
thin



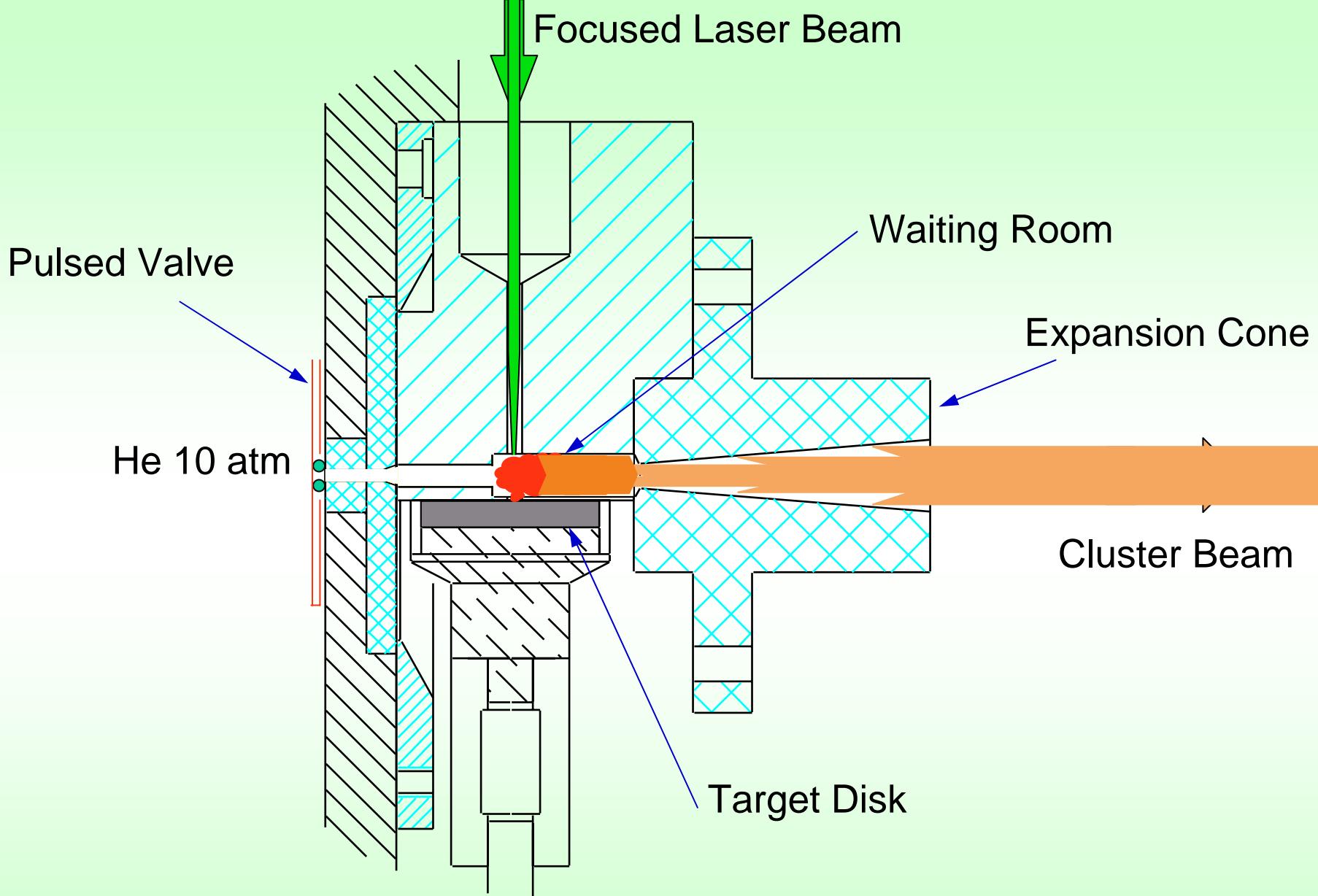
thin ← Mo → thick

Co/Mo [nm]
(I) 0.33/0.049
(II) 0.19/0.13
(III) 0.13/0.34

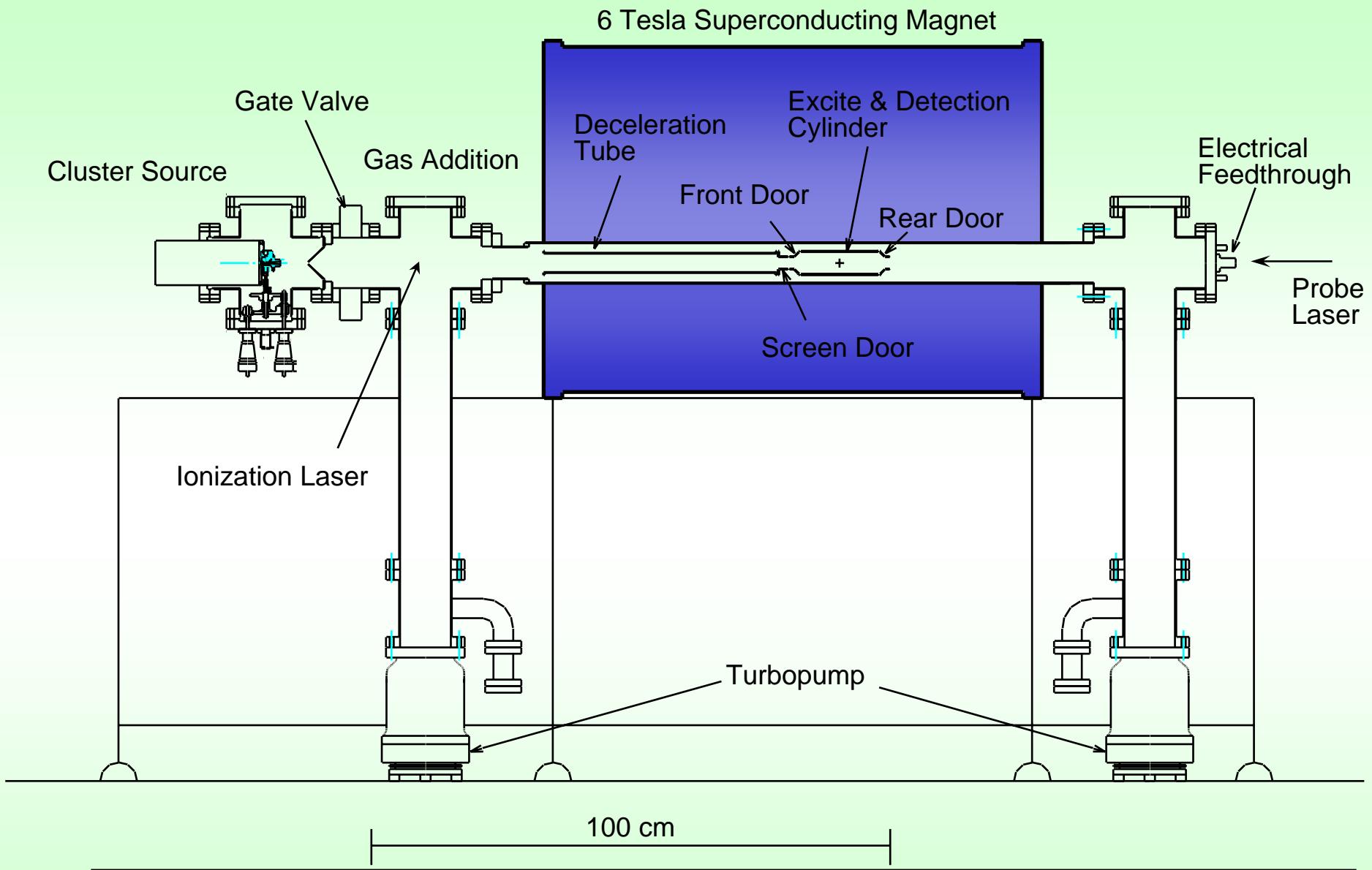
Reduction temperature : 800°C
Reaction temperature : 800°C
Pressure of ethanol : 30Torr
Reaction time : 10min



Initial Reaction on Catalyst

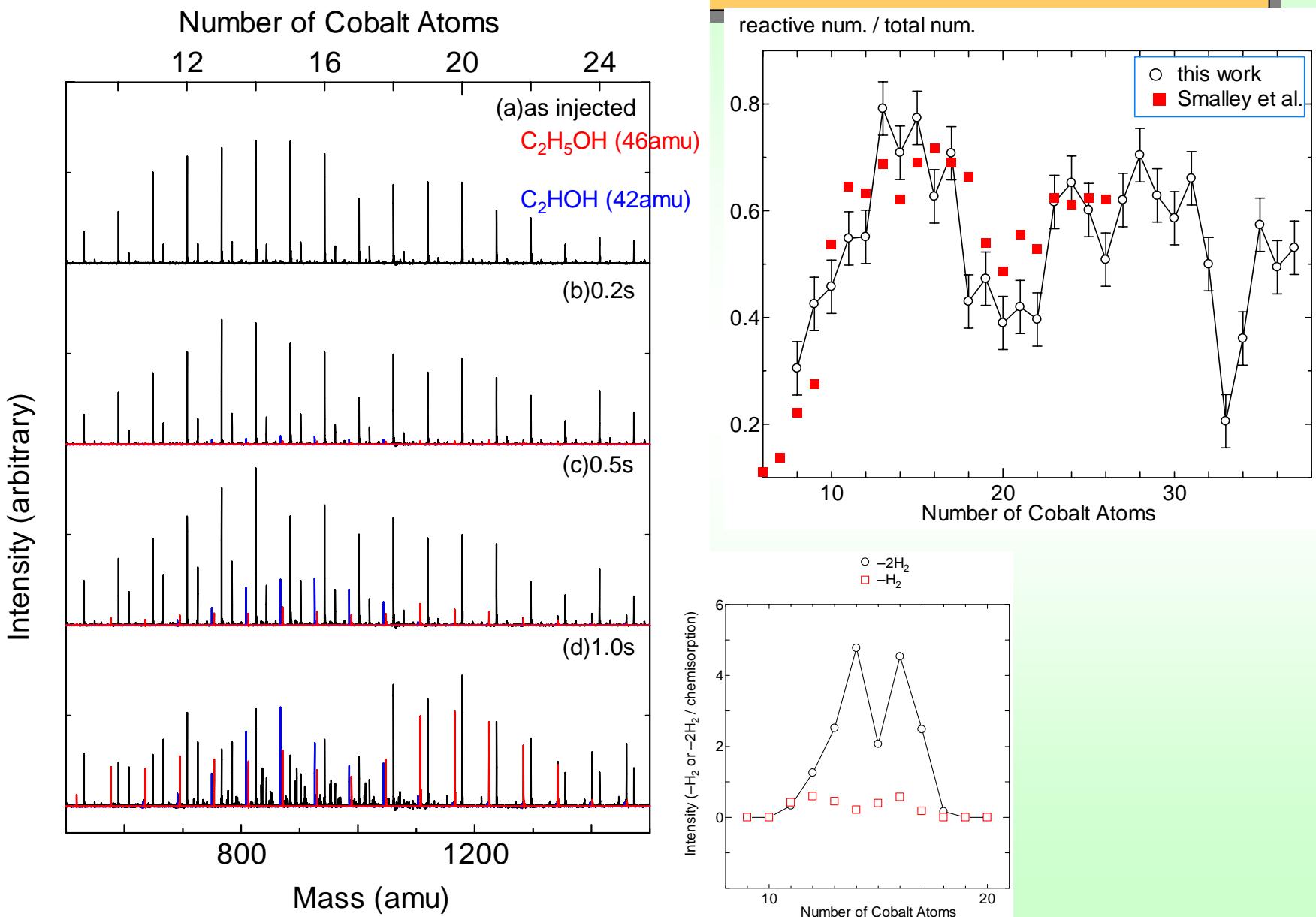


Cluster Source Nozzle for FT-ICR

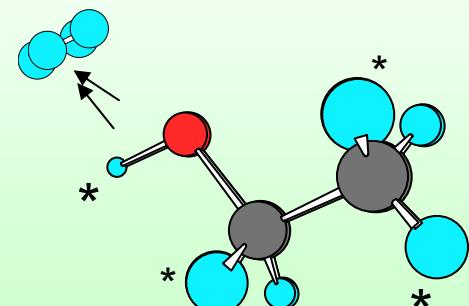
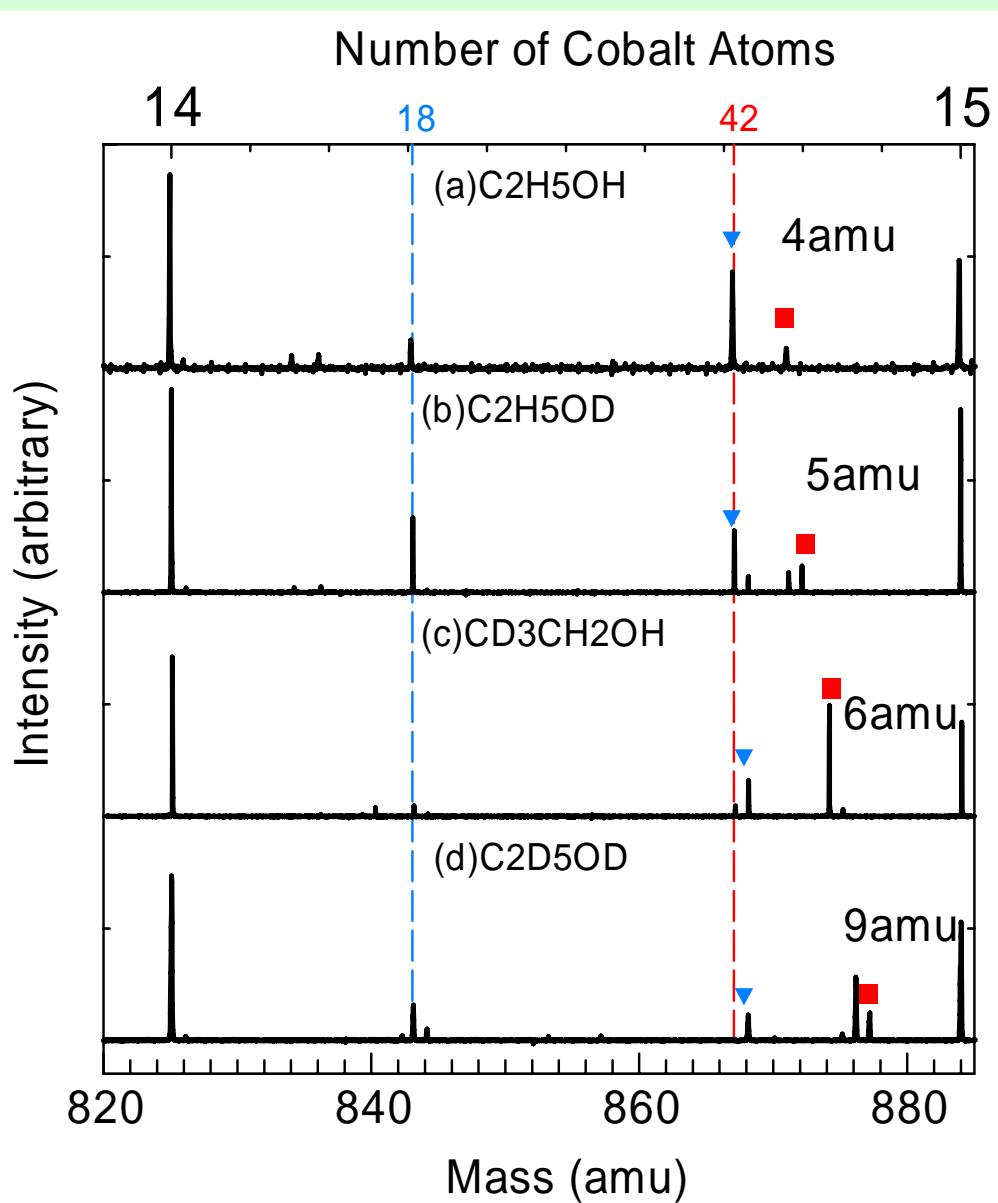


FT-ICR (Fourier Transform Ion Cyclotron Resonance)
Mass Spectrometer

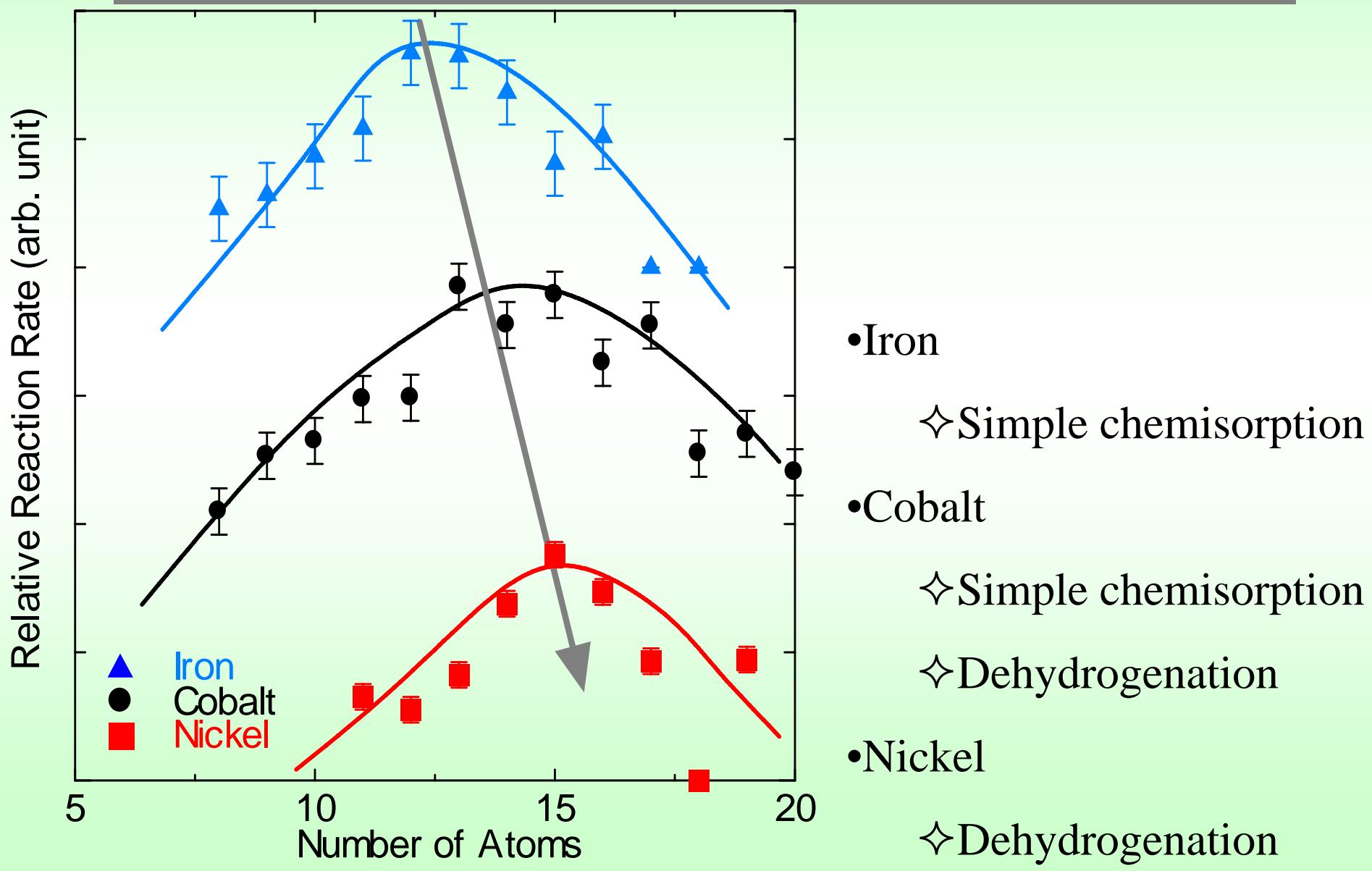
FT-ICR Study of Initial Reaction



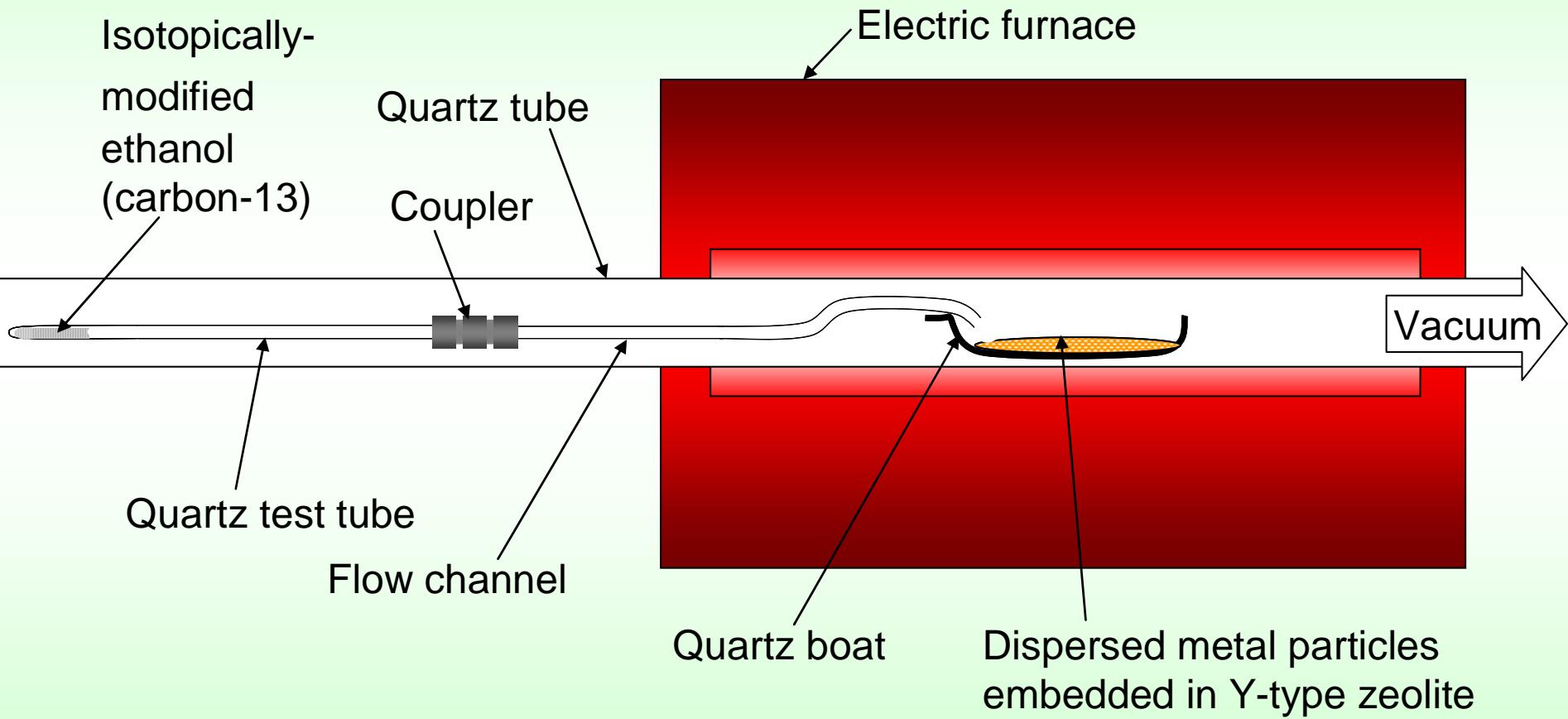
FT-ICR Chemical reaction of cobalt clusters (cation) with ethanol



FT-ICR Study of Initial Reaction

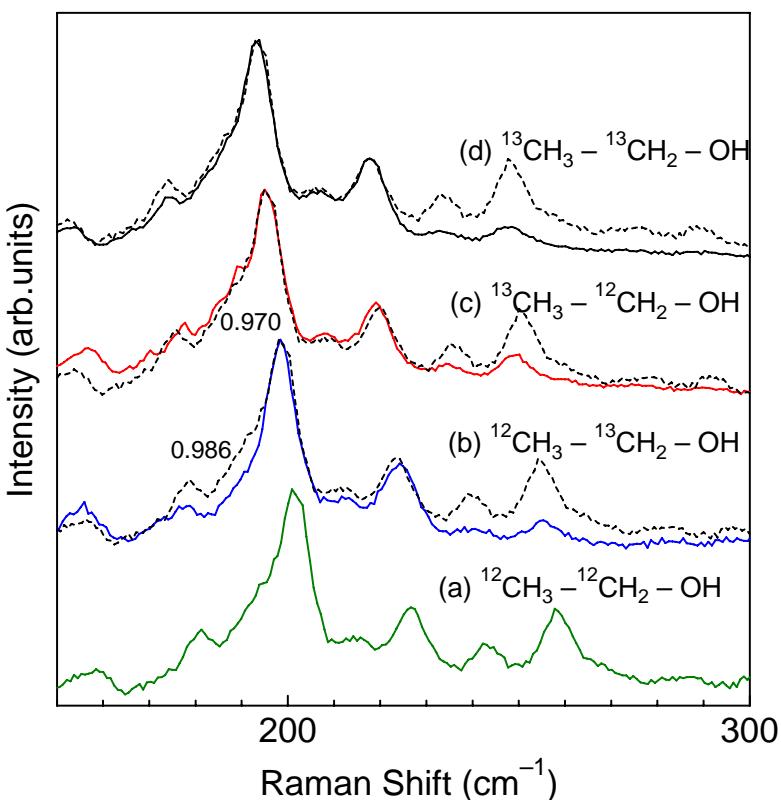
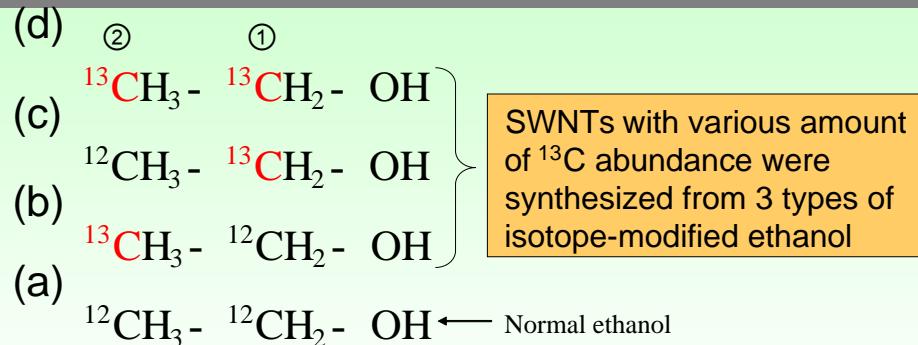


Single-Walled Carbon-13 Nanotubes (SW¹³CNTs)



ACCVD technique optimized for the efficient production of SWNTs from very small amount of ethanol

Comparison of Raman Spectra



Two carbon atoms in an ethanol molecule are not equally used for the SWNT formation in ACCVD process

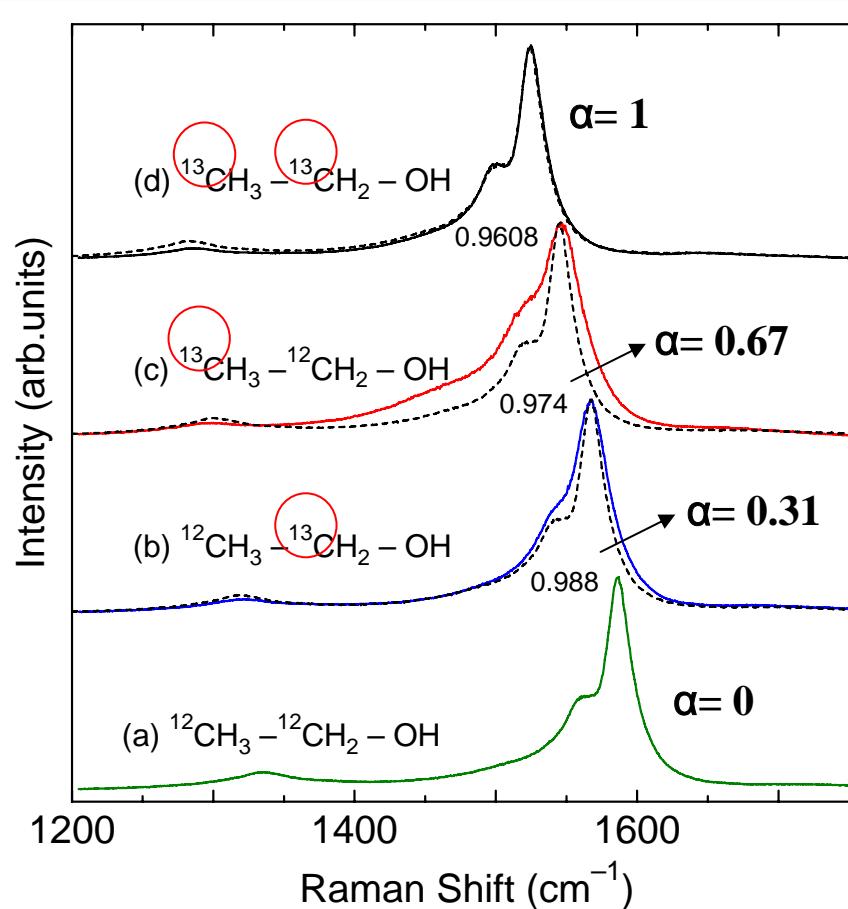
Assuming...

$$\nu_{\text{vib}} \propto \sqrt{\frac{1}{12}(1-\alpha) + \frac{1}{13}\alpha}$$

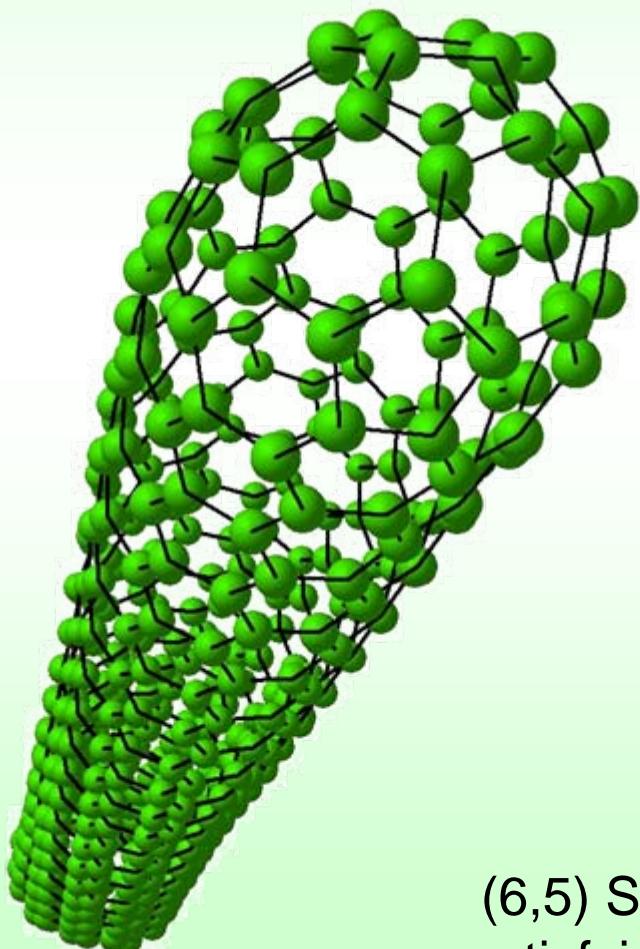
α : Ratio of ${}^{13}\text{C}$ in a SWNT

$$\nu_{\alpha} = \nu_{12\text{C}} \times \sqrt{(1-\alpha) + \frac{12}{13}\alpha}$$

Scaling factor



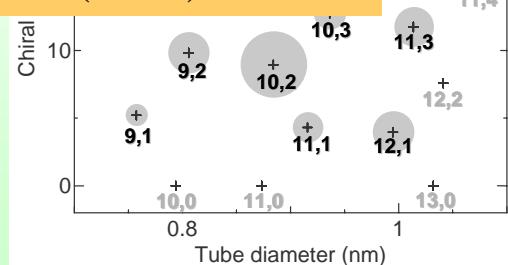
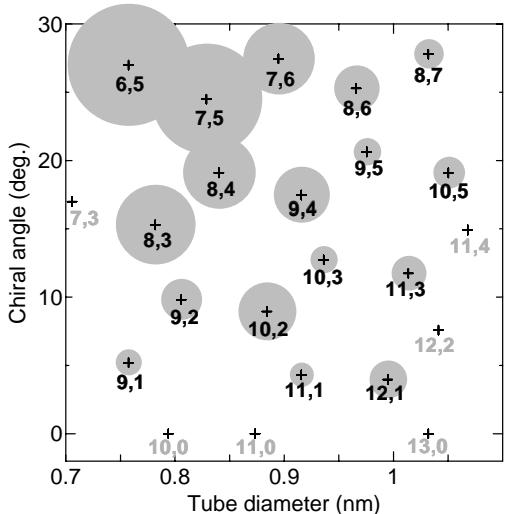
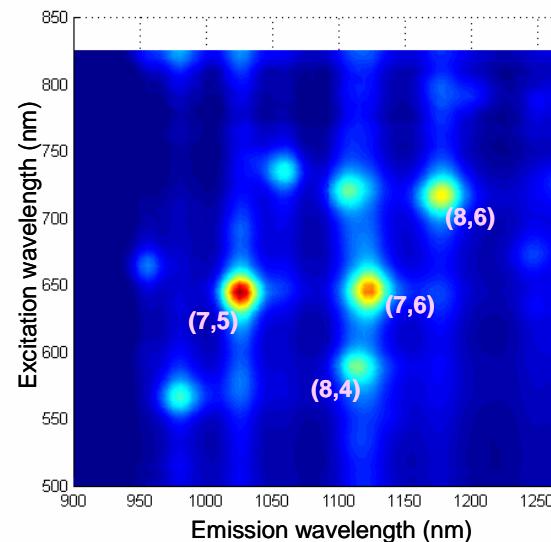
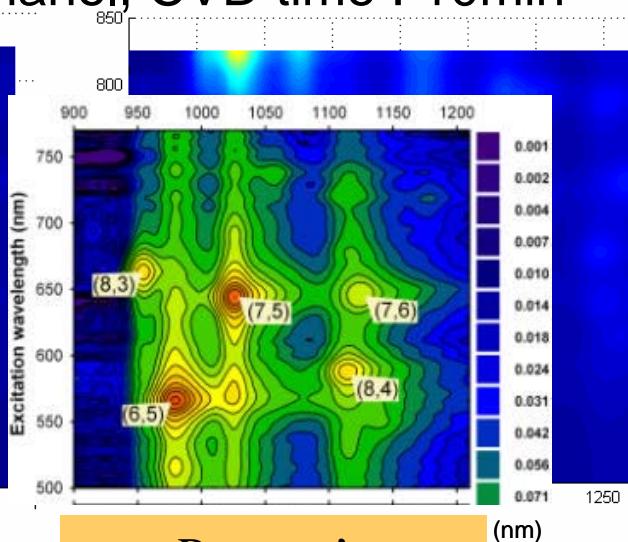
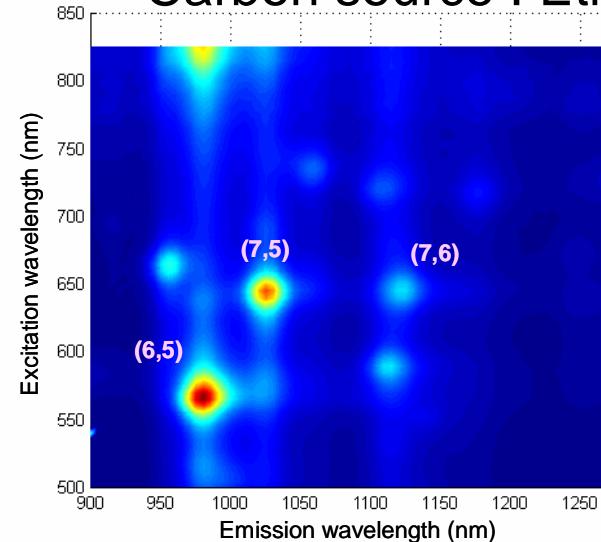
Determination of Chirality



(6,5) SWNT with unique cap structure
satisfying Isolated Pentagon Rule (IPR)

CVD Temperature Dependence

Carbon source : Ethanol, CVD time : 10min

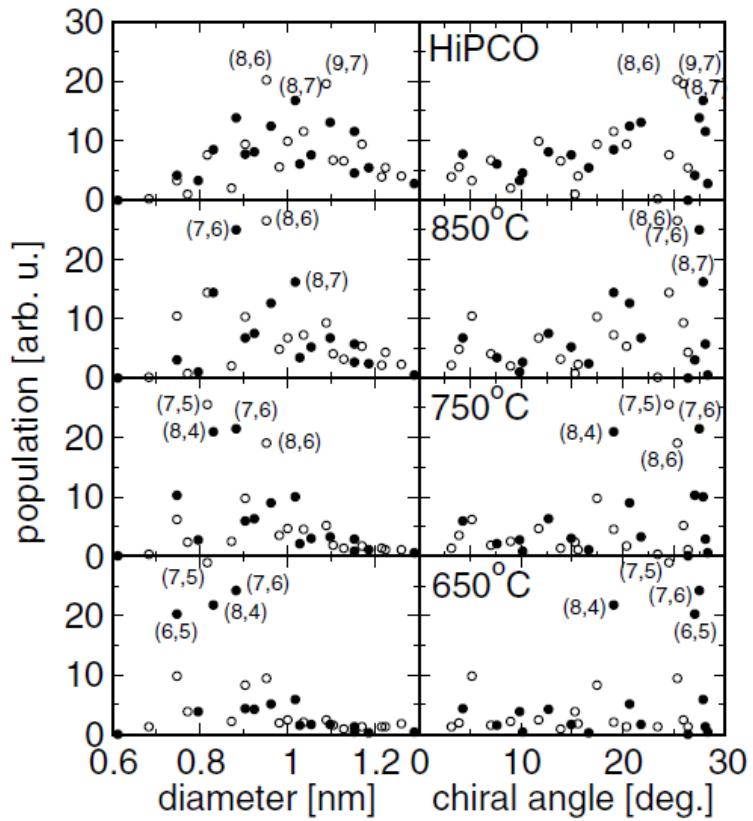
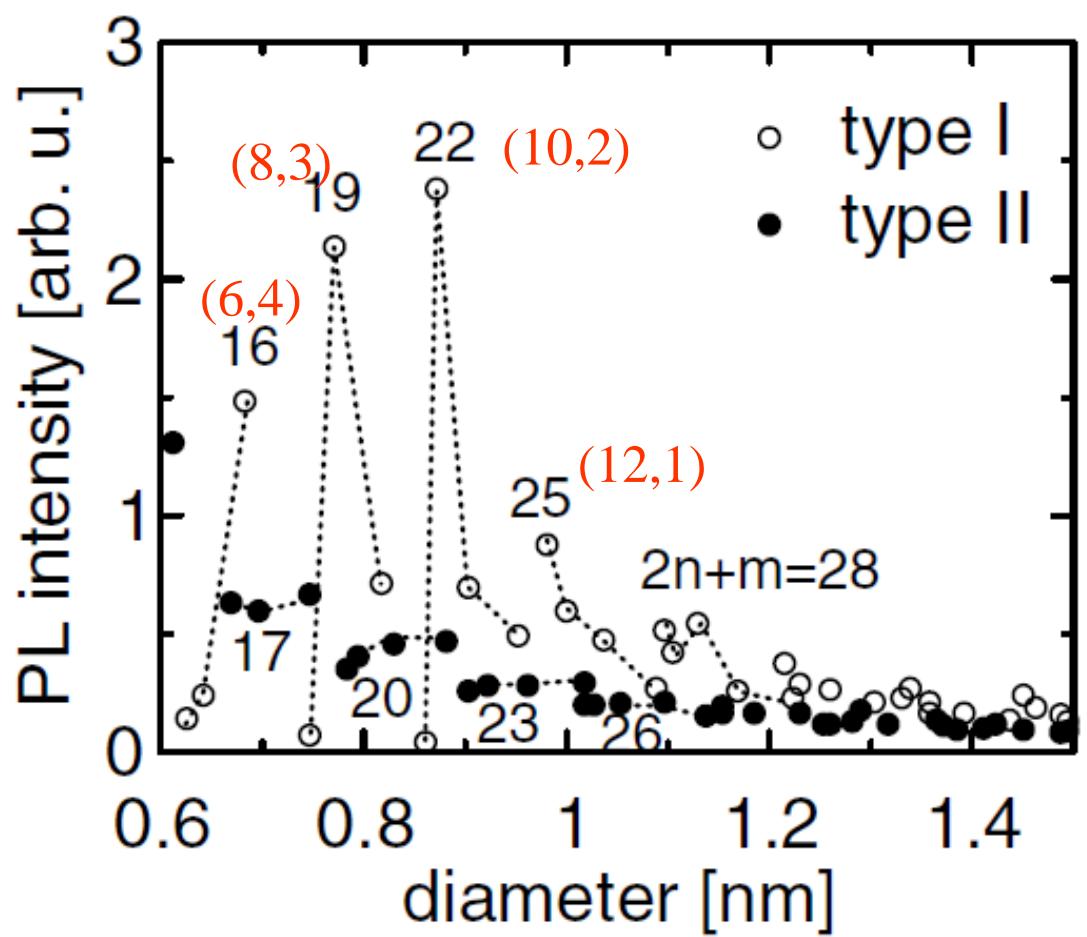


650°C

750°C

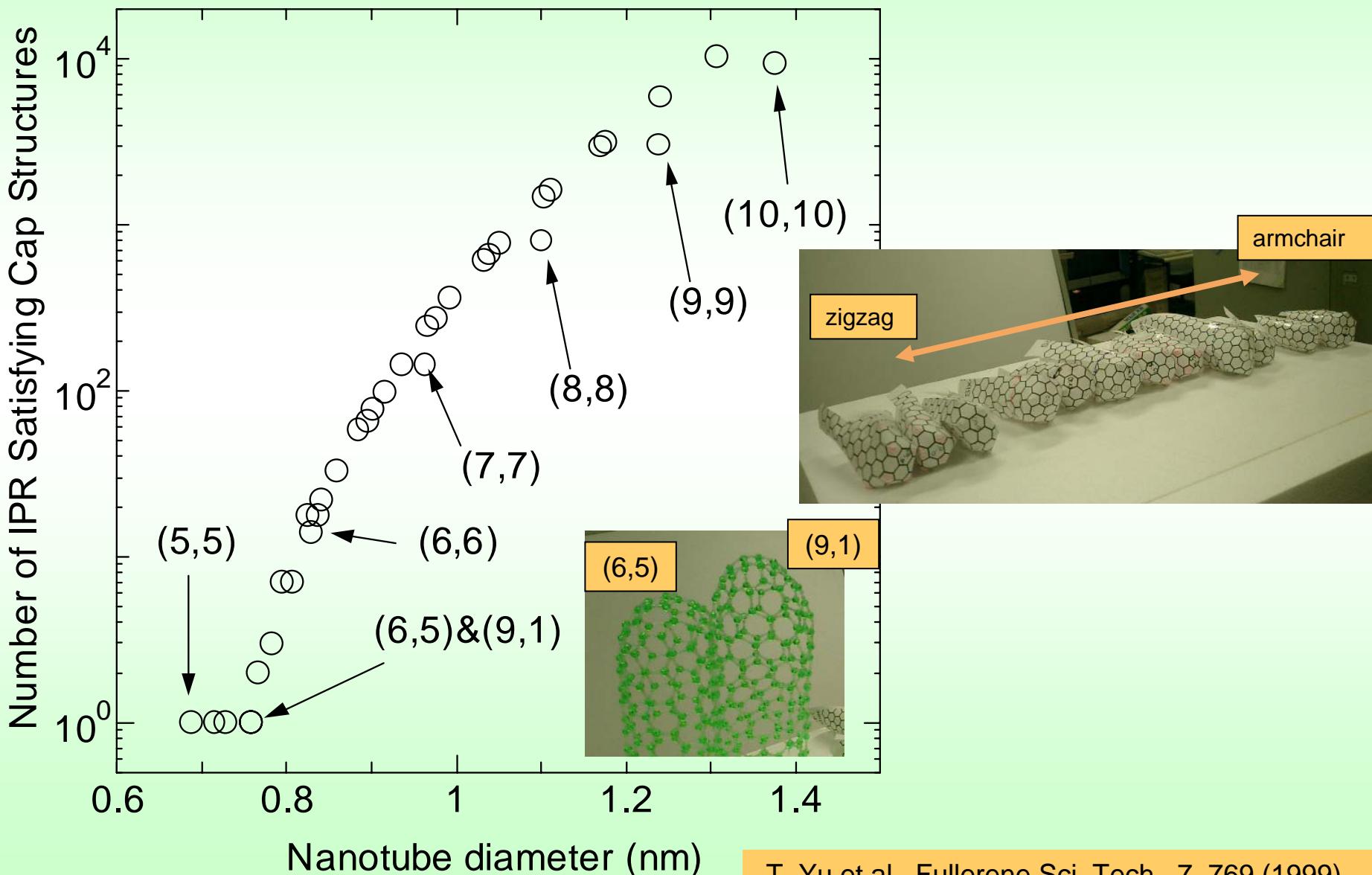
850°C

Chirality dependent quantum yield

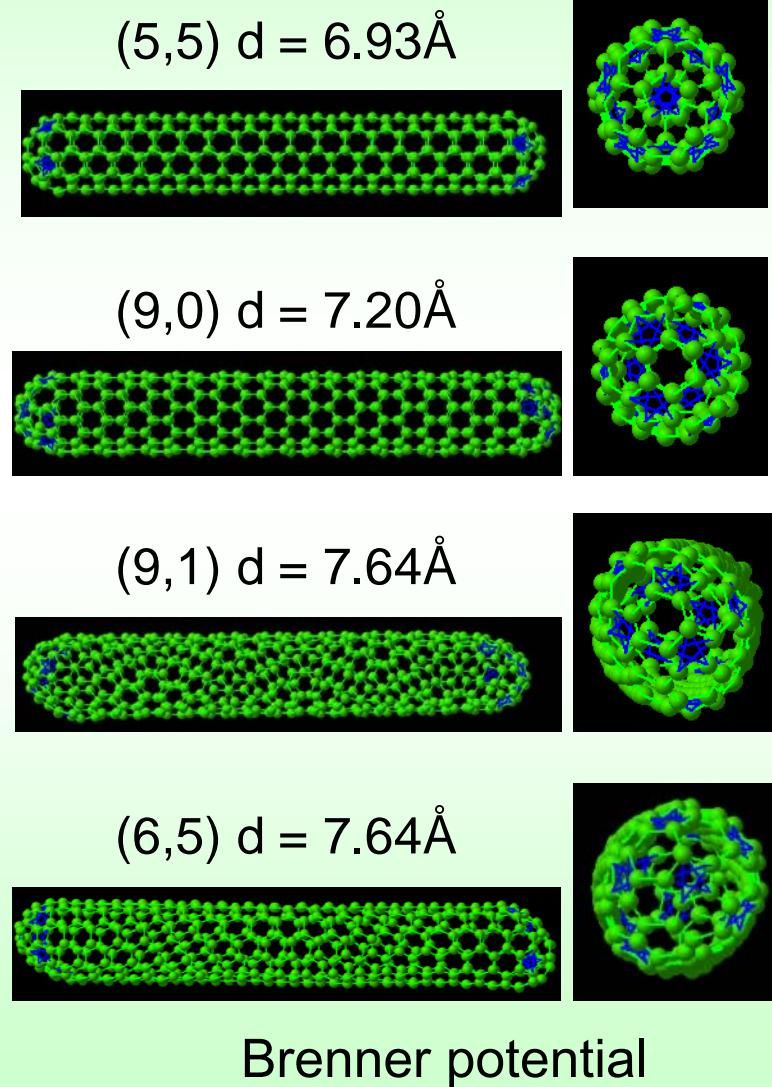
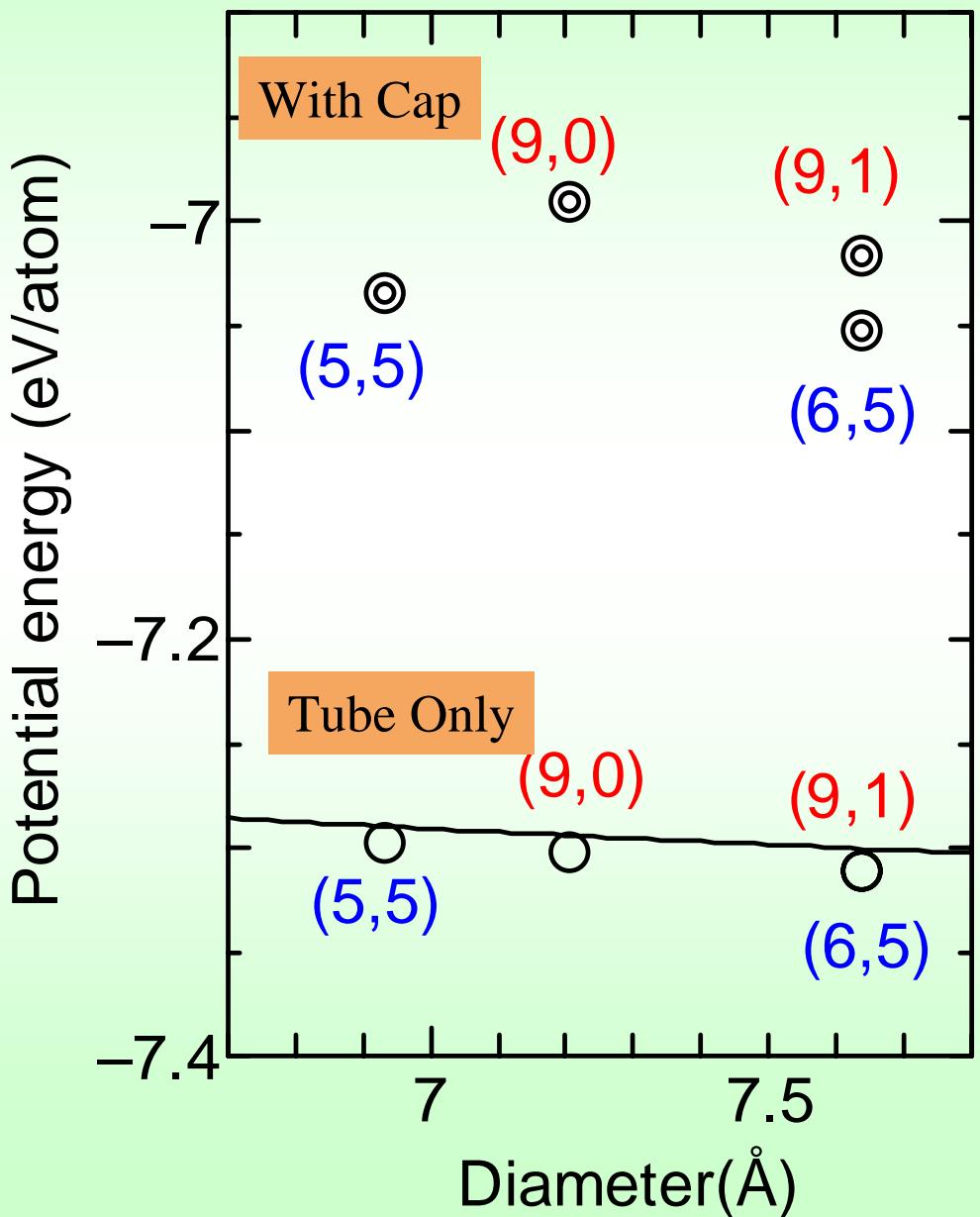


Y. Oyama, R. Saito, K. Sato, J. Jiang, Ge.G. Samsonidze, A. Gruneis, Y. Miyauchi, S. Maruyama, A. Jorio, G. Dresselhaus, M.S. Dresselhaus, Carbon in press.

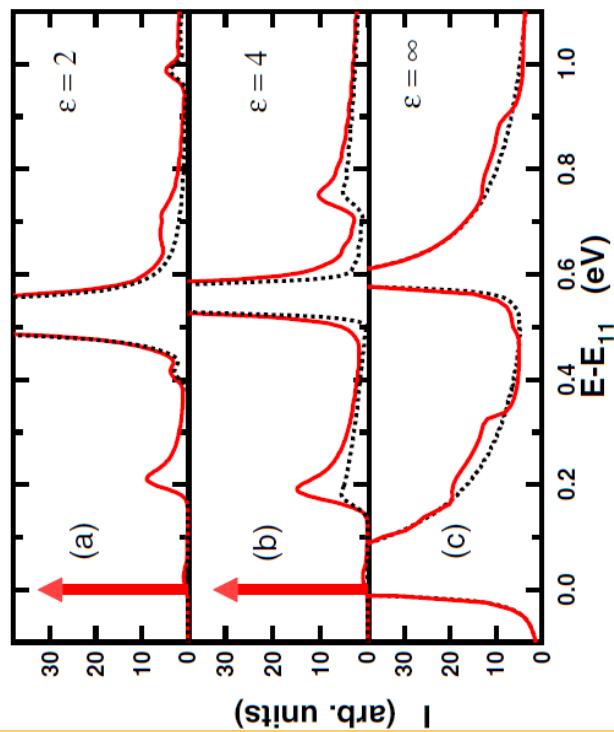
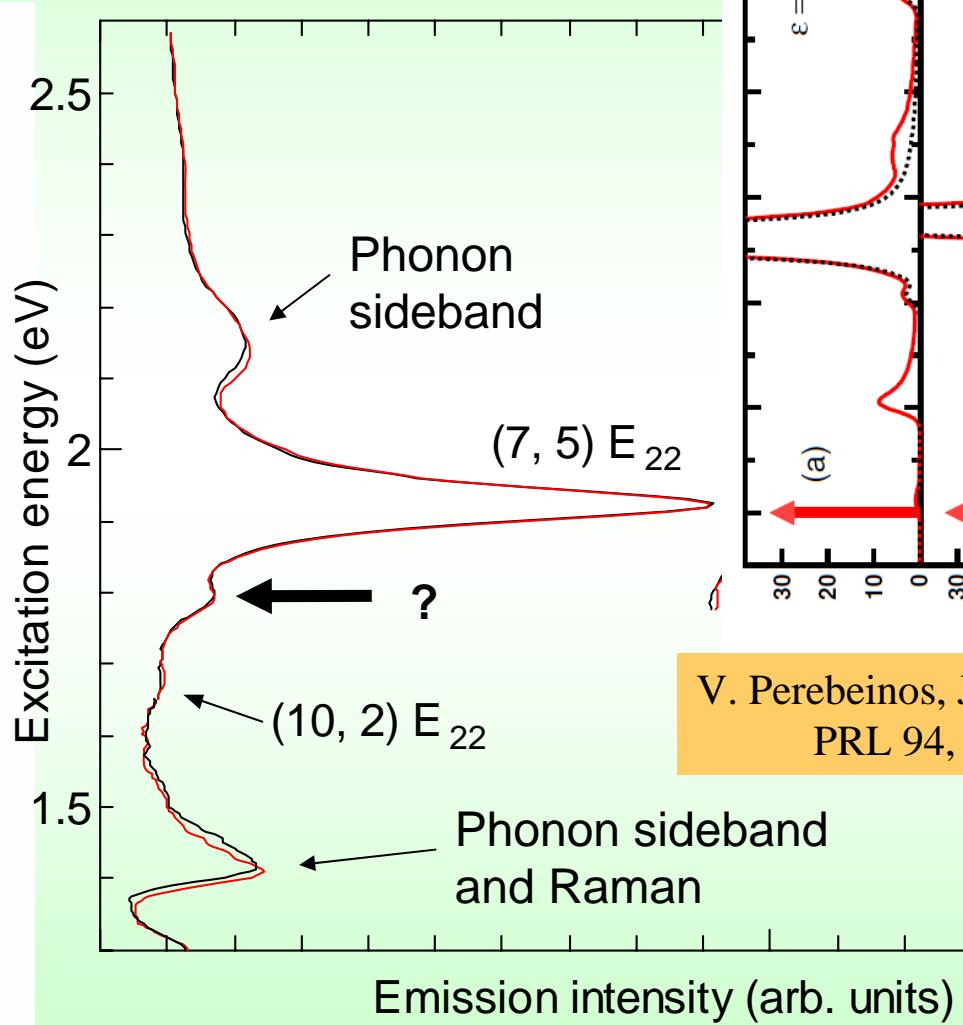
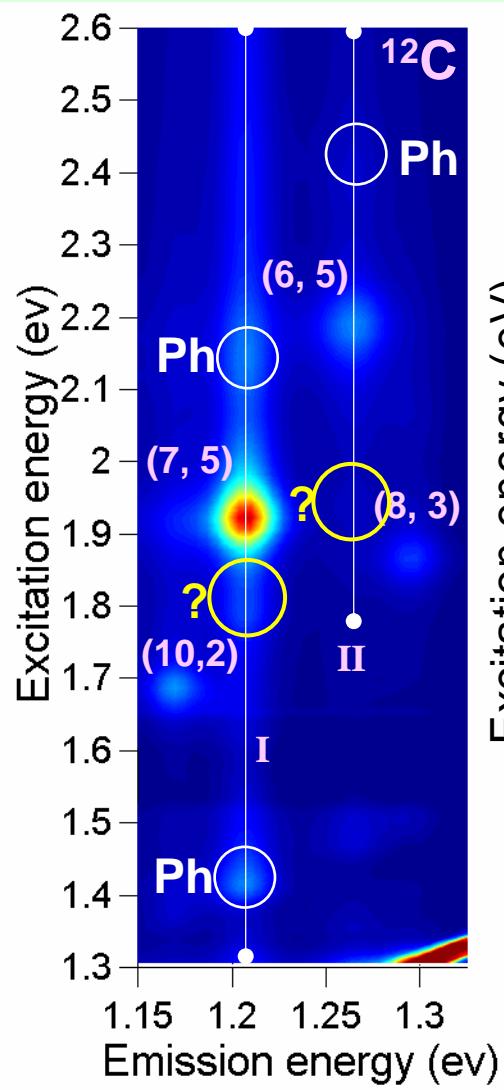
IPR (Isolated Pentagon Rule)-Satisfying Cap Structure



Energy of Nanotube Cap Structure



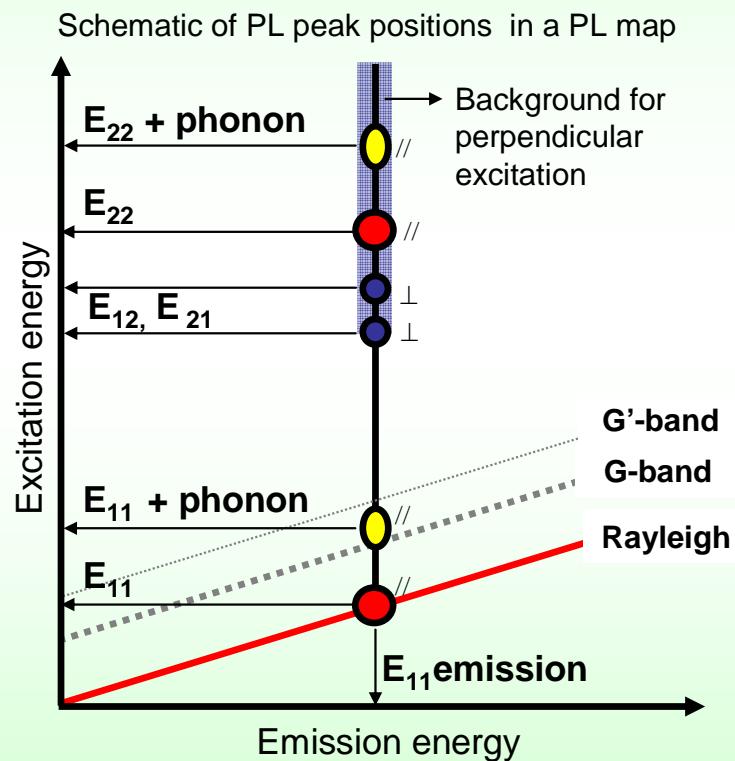
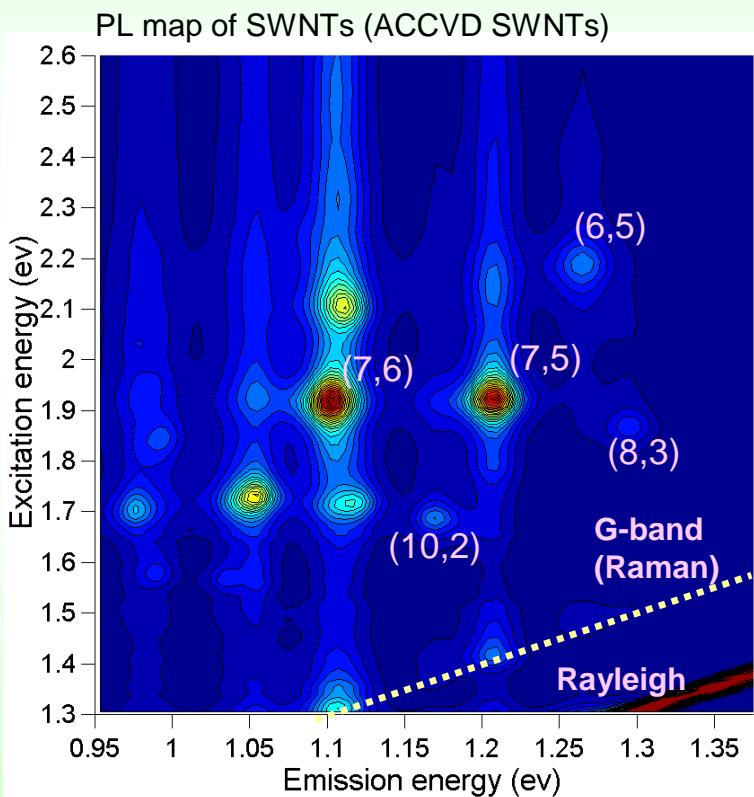
PL peaks other than E_{ii} ~ Phonon Sideband



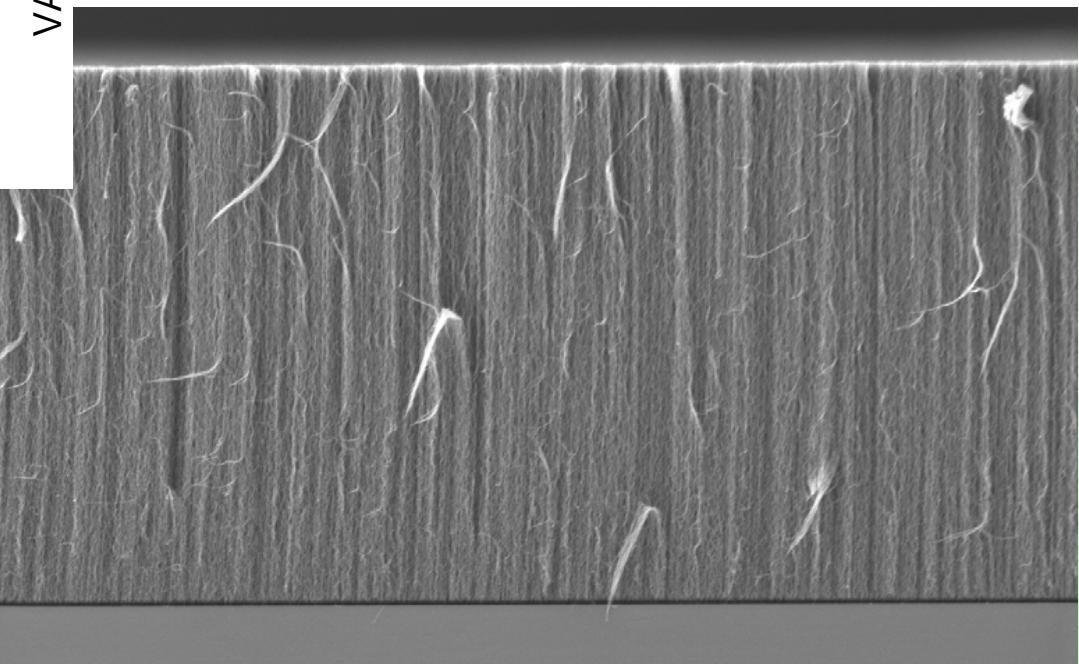
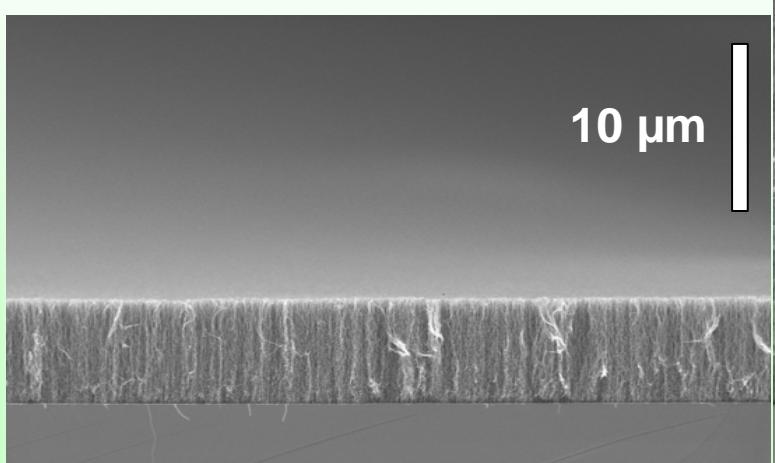
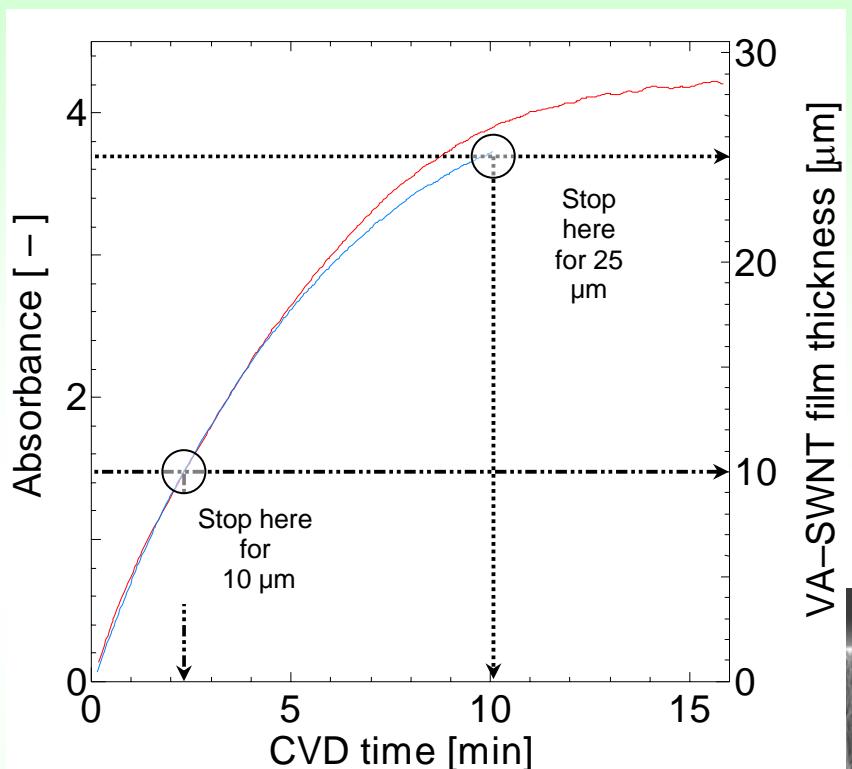
V. Perebeinos, J. Tersoff, P. Avouris,
PRL 94, 027402 (2005).

— ^{12}C
— ^{13}C

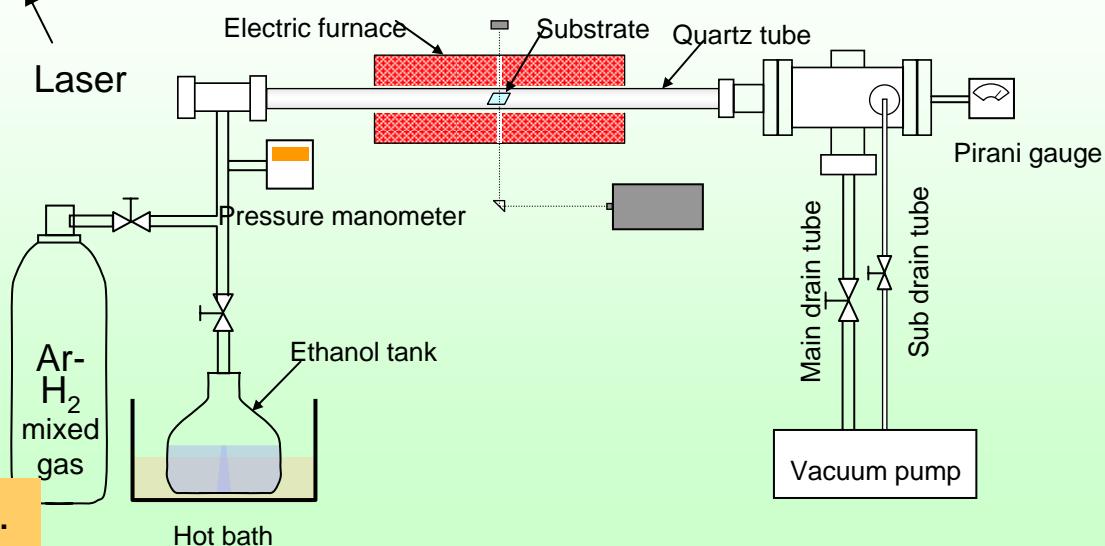
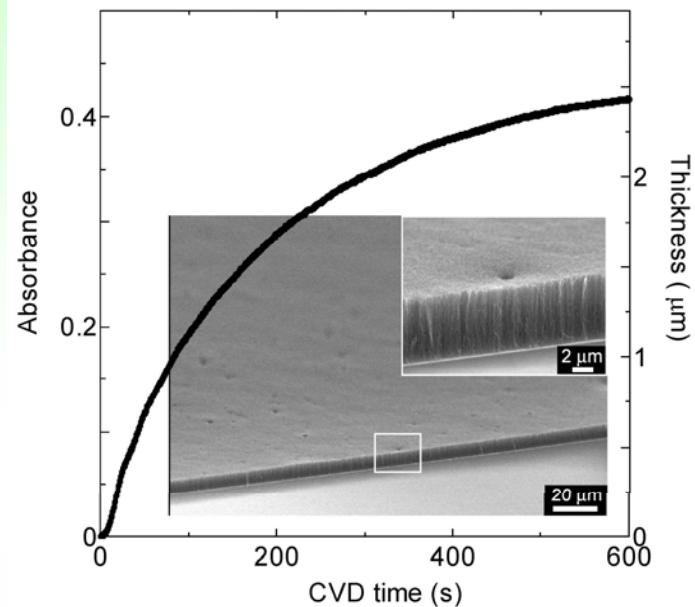
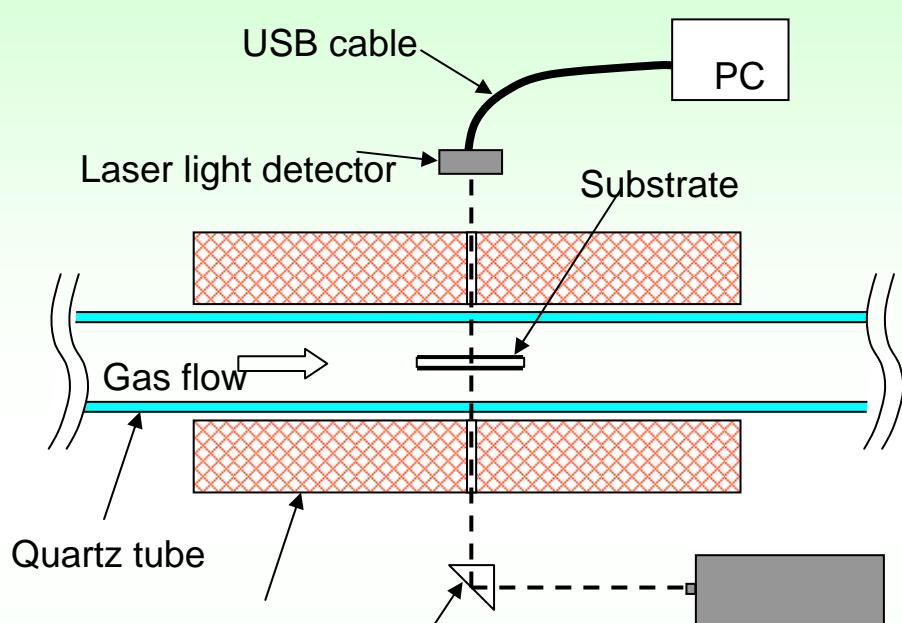
PL peaks other than E_{ii}



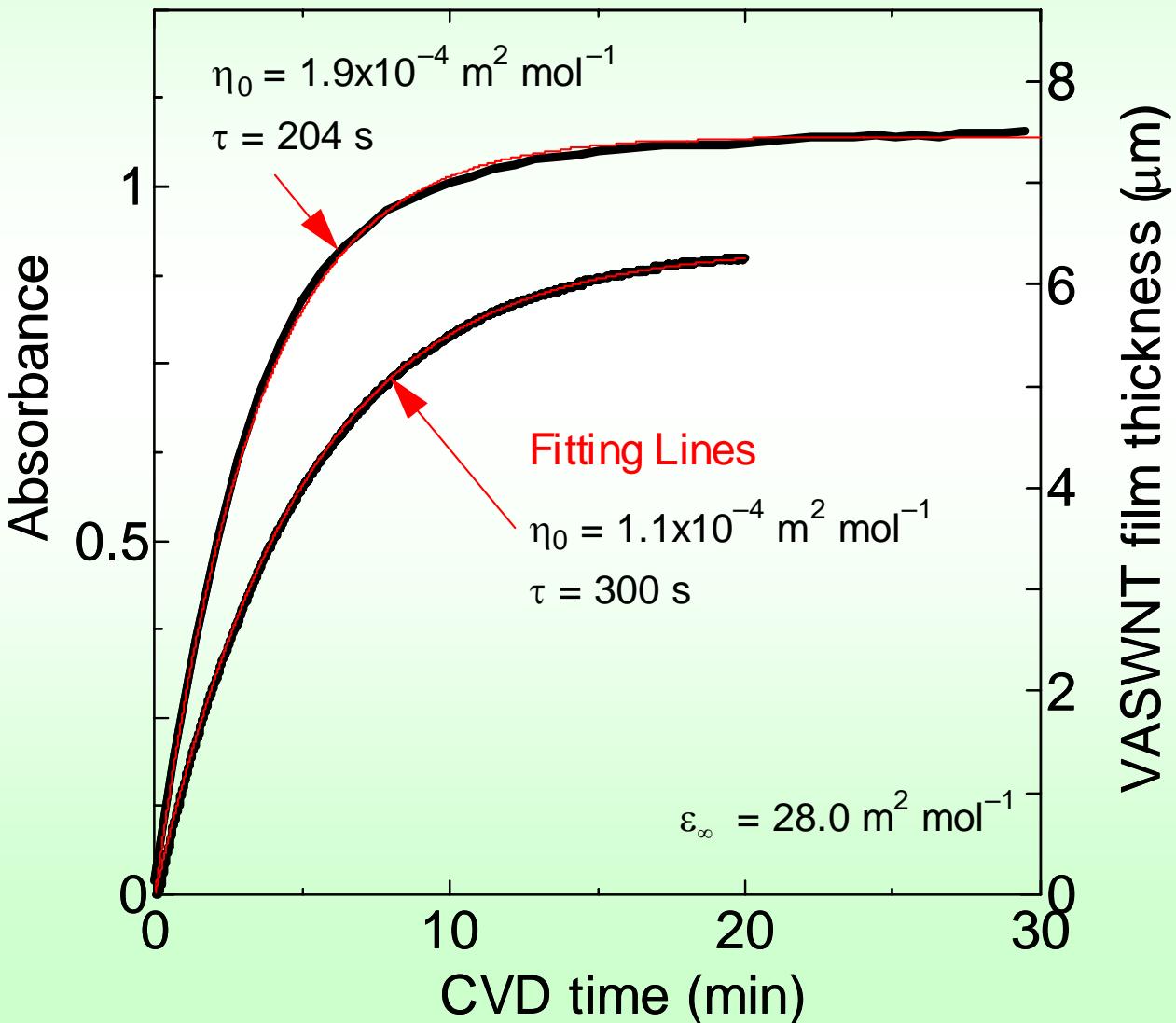
Vertically Aligned SWNTs and Growth Process



In-situ Measurement of Film Thickness



Catalyst-Consumption Growth Model



Growth Model

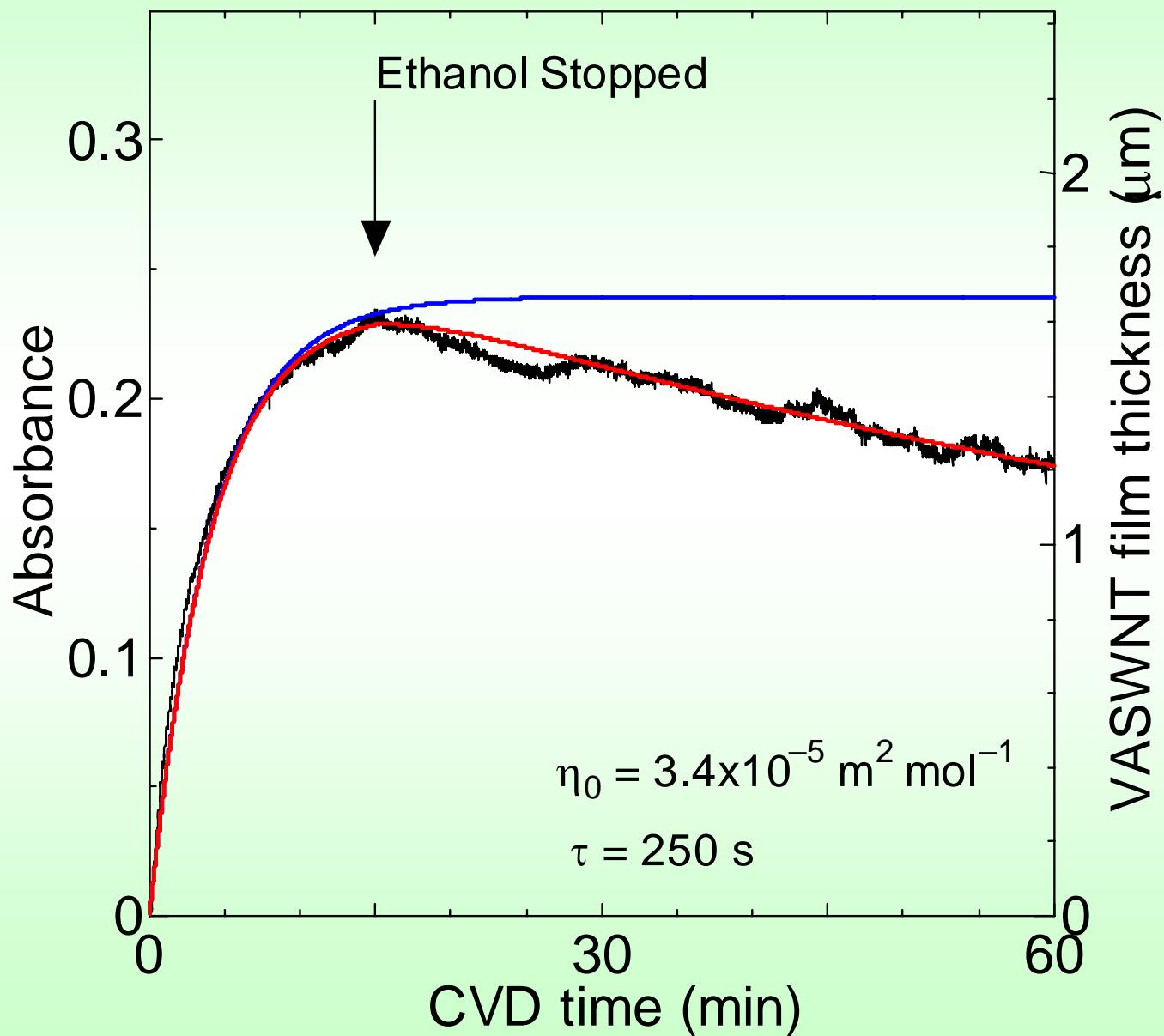
Catalyst Activity η

$$\frac{d\eta}{dt} = -\tau\eta$$

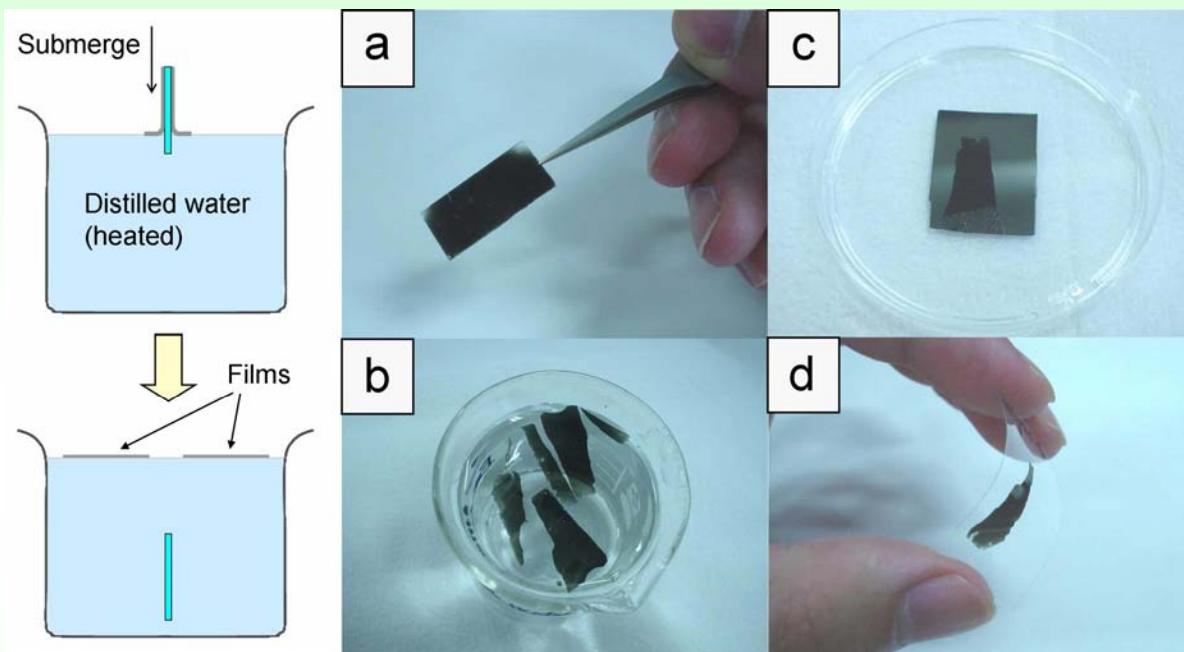
Carbon Mass $C(t)$

$$C(t) = \eta_0 \tau \left(1 - e^{-t/\tau}\right)$$

Additional Burning of SWNTs



Detachment of VA-SWNT films with hot water



(a) As-grown on quartz

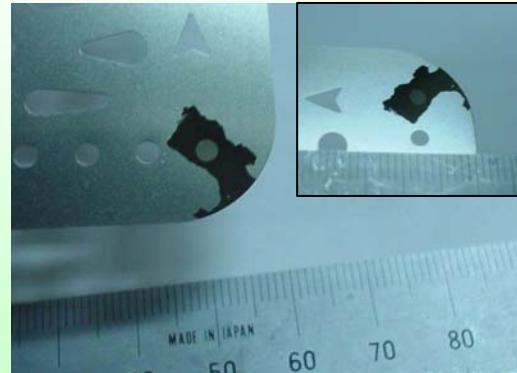
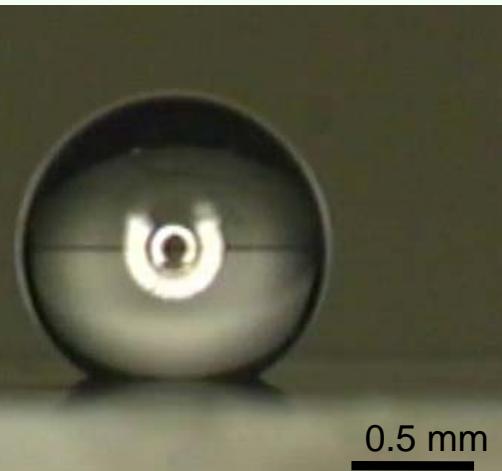
S4700-73 1.0kV 5.2mm x3.00k SE(M) 05/10/27

10 μm
10.0um

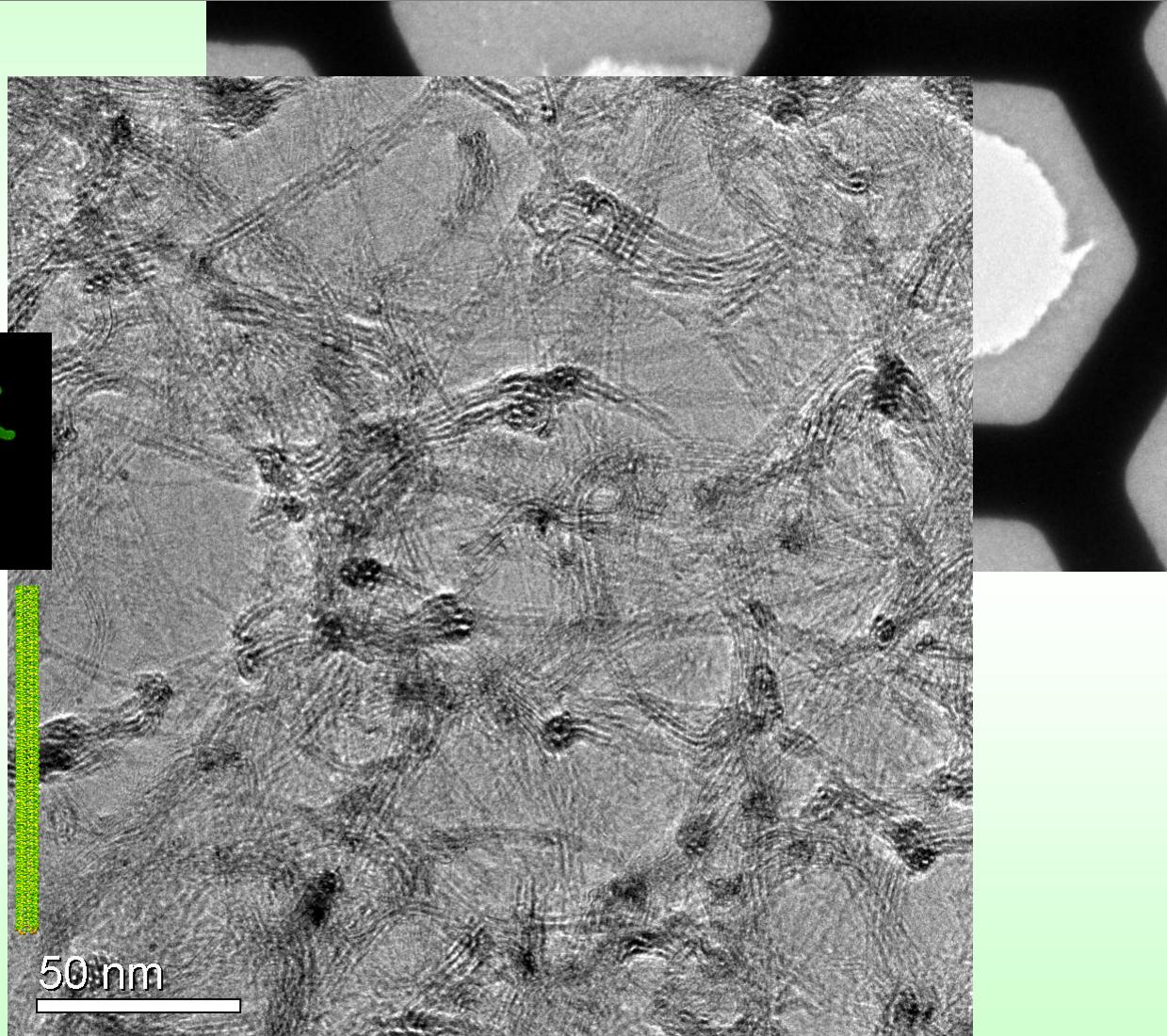
(b) Re-attached on Si

S4700-62 1.0kV 4.9mm x3.00k SE(M) 05/10/27

10 μm
10.0um

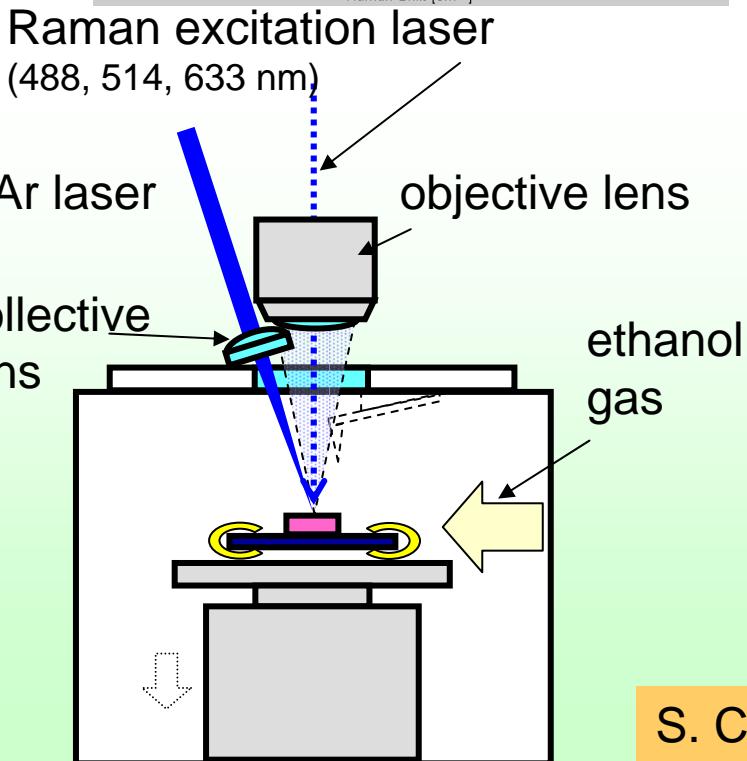
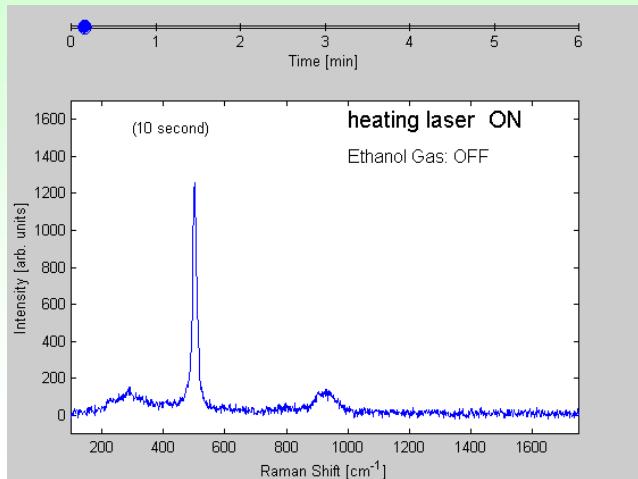


TEM Image from Top of Carpet

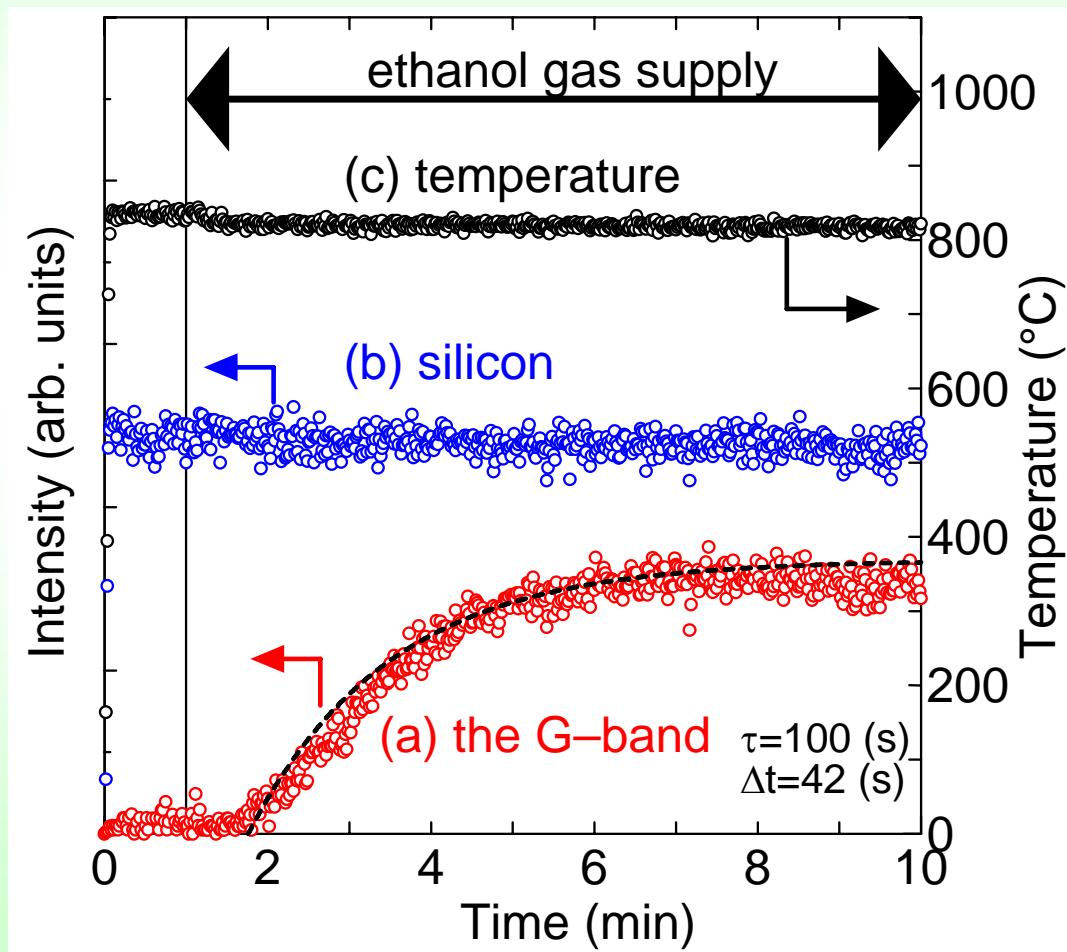


Hidetsugu Shiozawa & Thomas Pichler (IFW-Dresden e.V.),
Erik Einarsson, Shigeo Maruyama

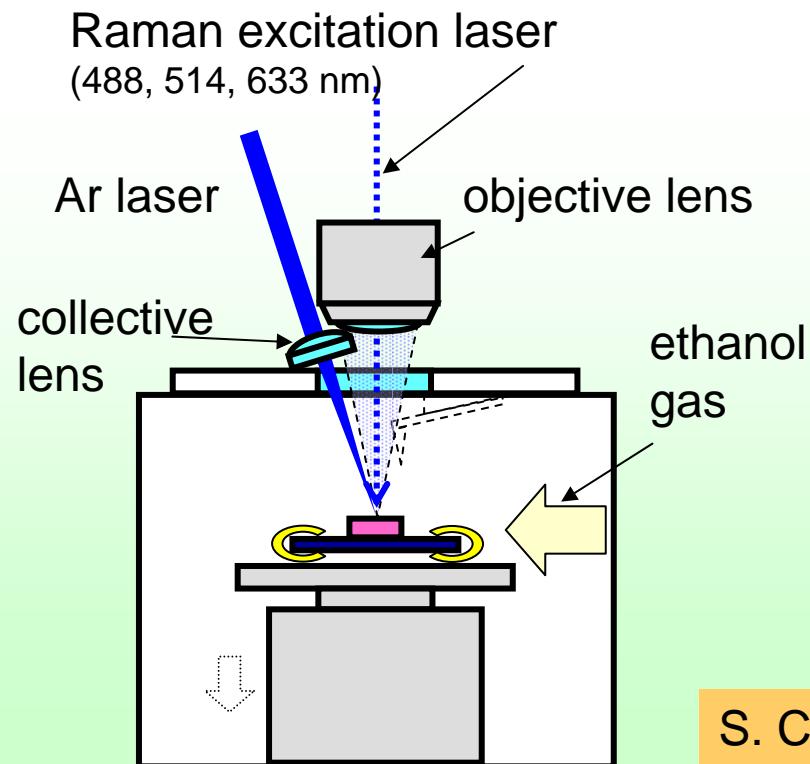
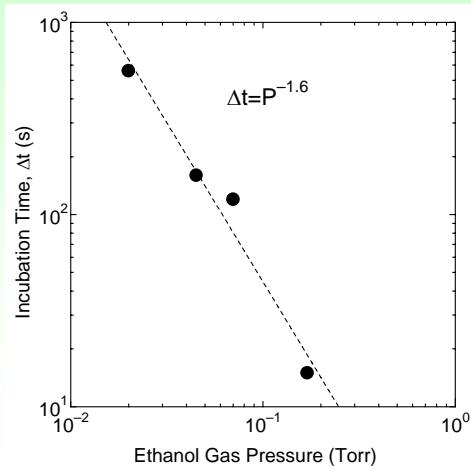
Laser ACCVD and in situ Raman Measurements



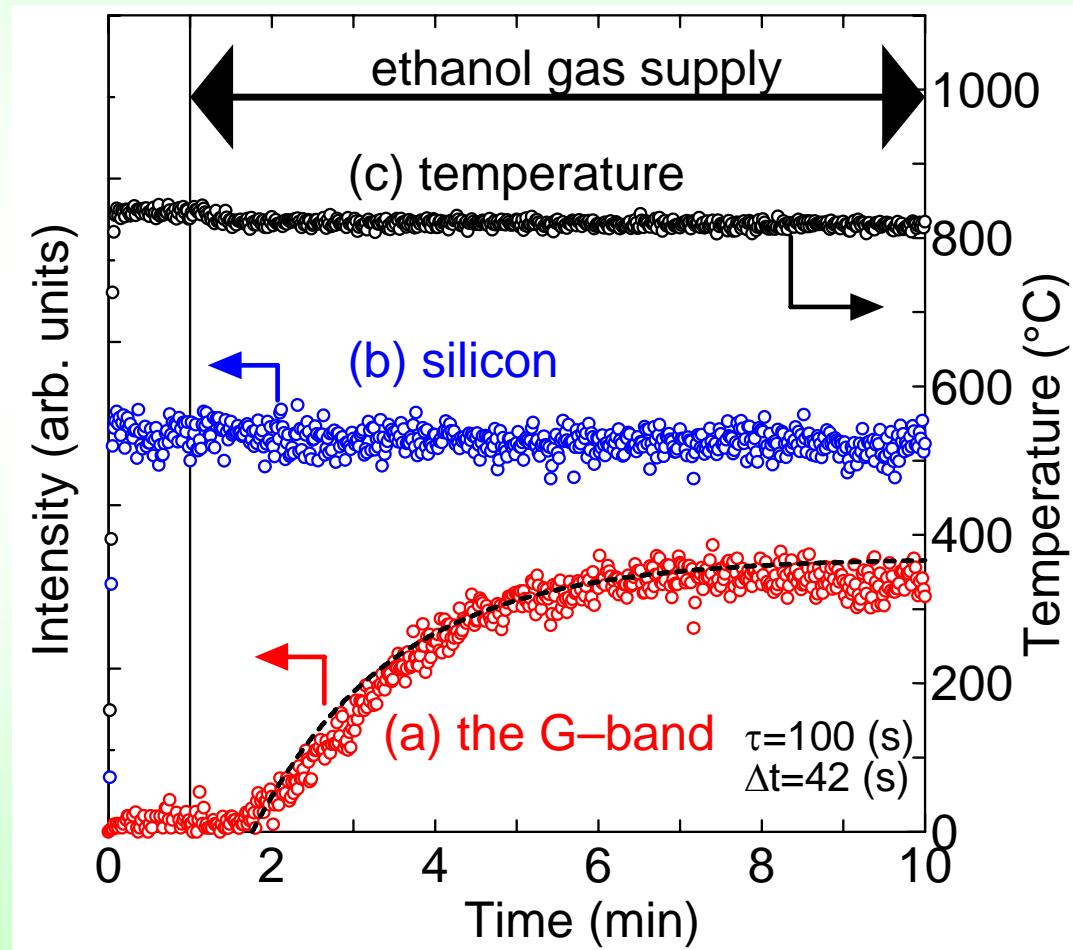
$$y = A \left[1 - \exp\left(\frac{t - \Delta t}{\tau}\right) \right] \quad \tau = 100 \sim 200 \text{ (s)} \\ \Delta t = 30 \sim 40 \text{ (s)}$$



Laser ACCVD and in situ Raman Measurements



$$y = A \left[1 - \exp\left(\frac{t - \Delta t}{\tau}\right) \right] \quad \tau = 100 \sim 200 \text{ (s)} \\ \Delta t = 30 \sim 40 \text{ (s)}$$



Acknowledgements

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Prof. K. Matsuda @ Kyoto University,

Prof. R. Saito @ Tohoku Univ., Prof. S. Okada @ Tsukuba Univ., Prof. T. Miyake @ AIST

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Mr. M. Sunose @ Seki-Tech.

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Mr. S. Watanabe, Ms. M. Nakagawa, Mr. R. Tamochi @ Hitachi Science Systems

JSPS, MEXT, Toray, XNRI, J-Power, DENSO,
Konica-Minolta, Tokyo-Gas, SEL, Toso

Professor Richard E. Smalley at Rice University (1943-2005)

1943: Born in Akron, Ohio on June 6

1976: PhD from Princeton,
Assistant Professor at Rice
Laser Vaporization Cluster Beam

1984: Discovery of C60

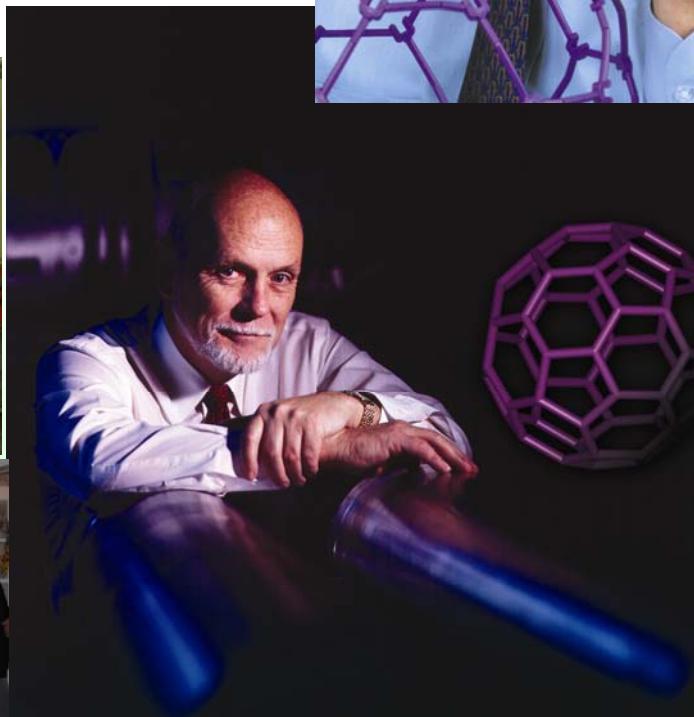
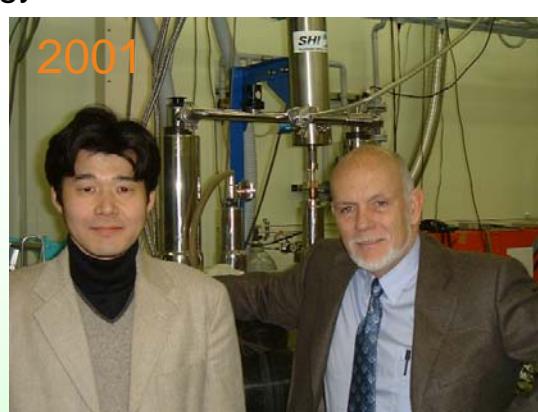
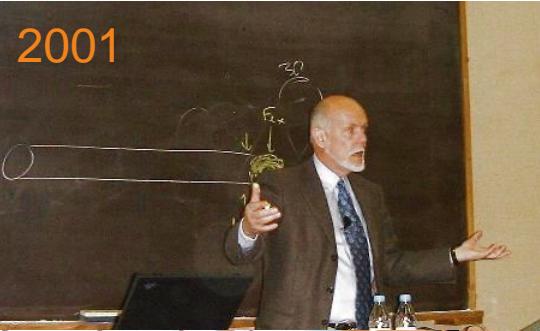
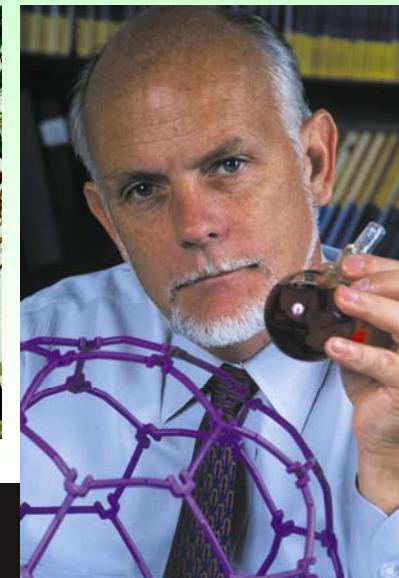
1996: Nobel Prize in Chemistry with Kroto and Curl

1996: Bulk Production of SWNTs
Then HiPco Process

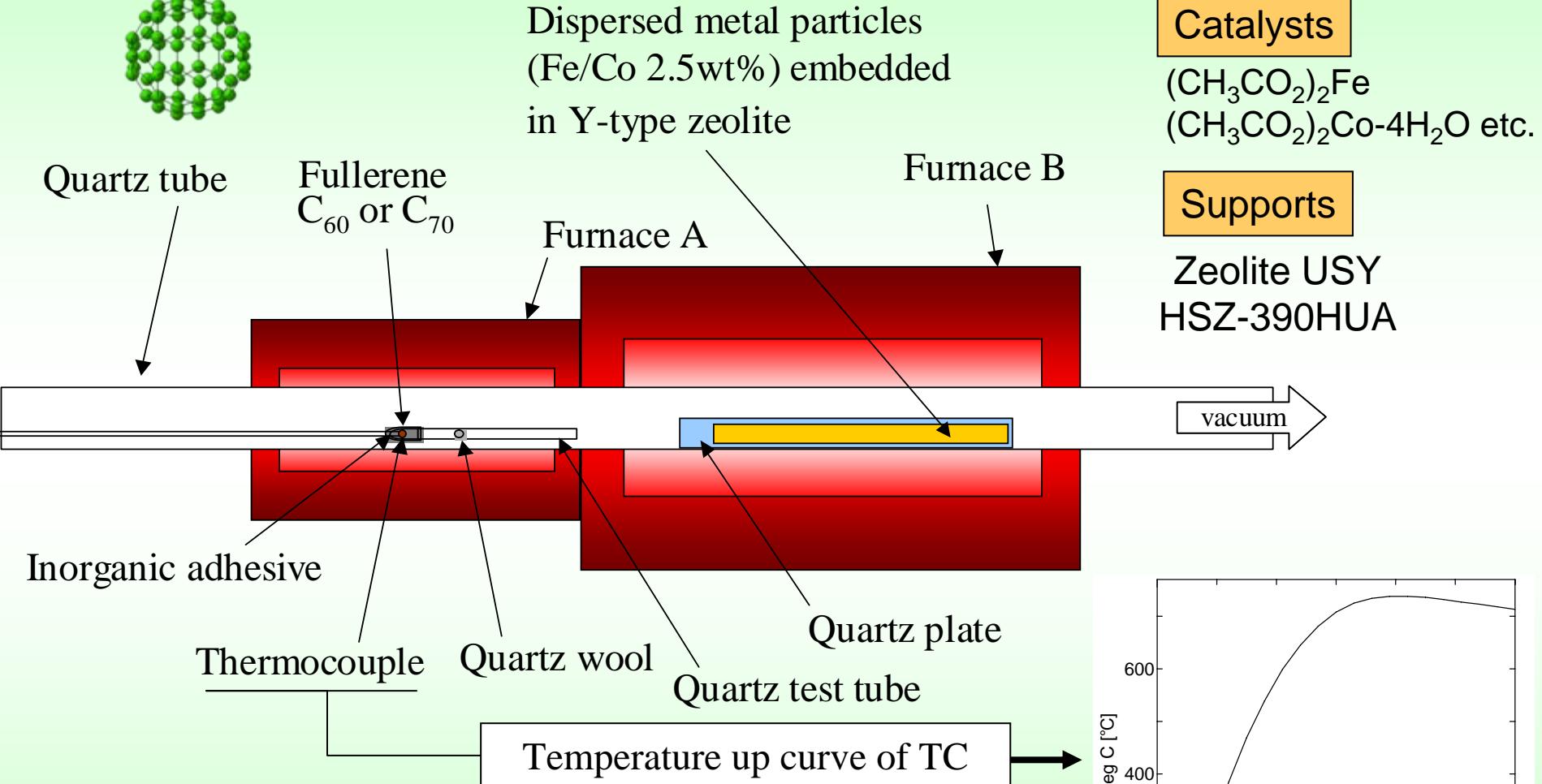
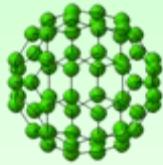
2000: Testified for National Nanotechnology Initiative

2000: Found Carbon Nanotechnology Inc.
Application of Nanotechnology to Energy Problem

2005: Died at 62 on October 28, 2005



Catalytic CVD Synthesis of SWNTs from Fullerene



Temperature: Furnace B 825°C

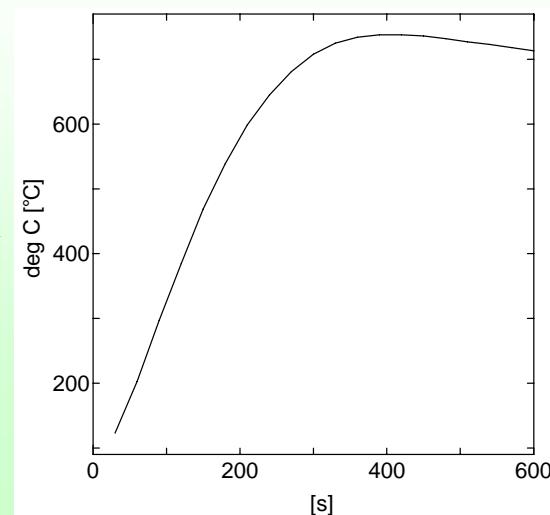
Time: 10min

Catalysts

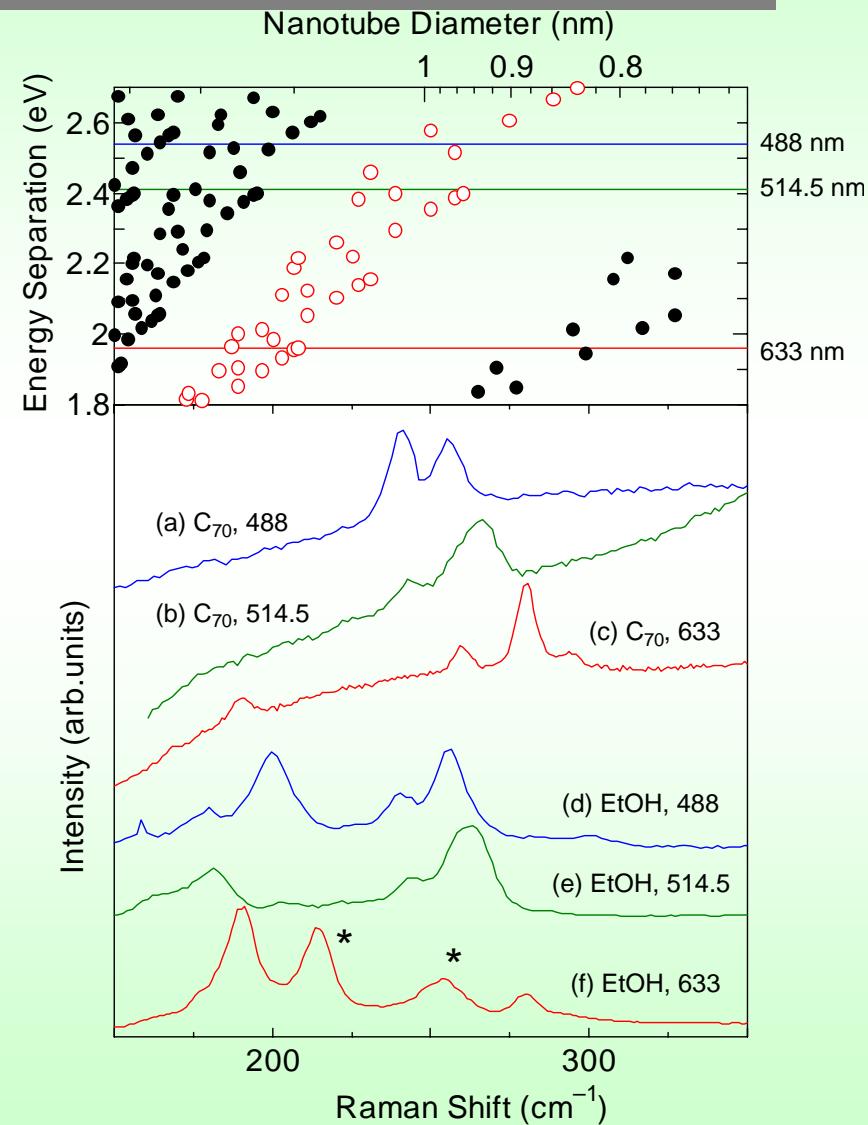
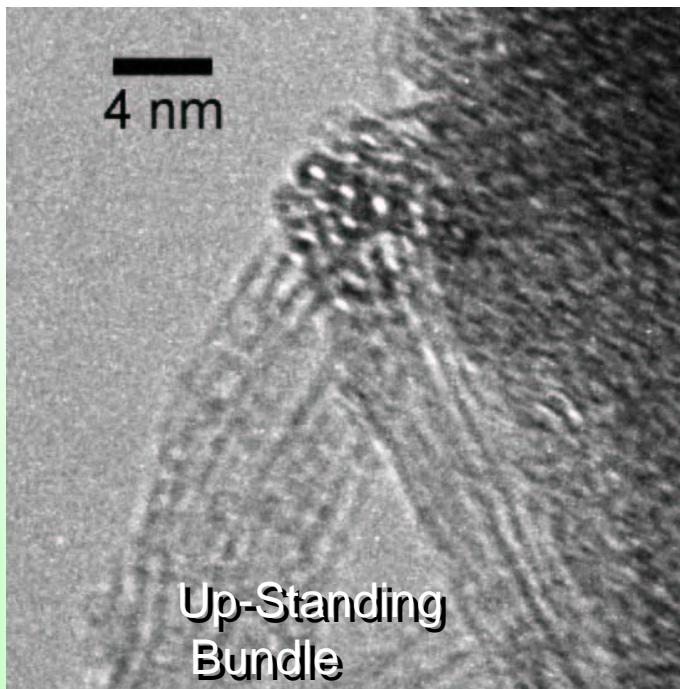
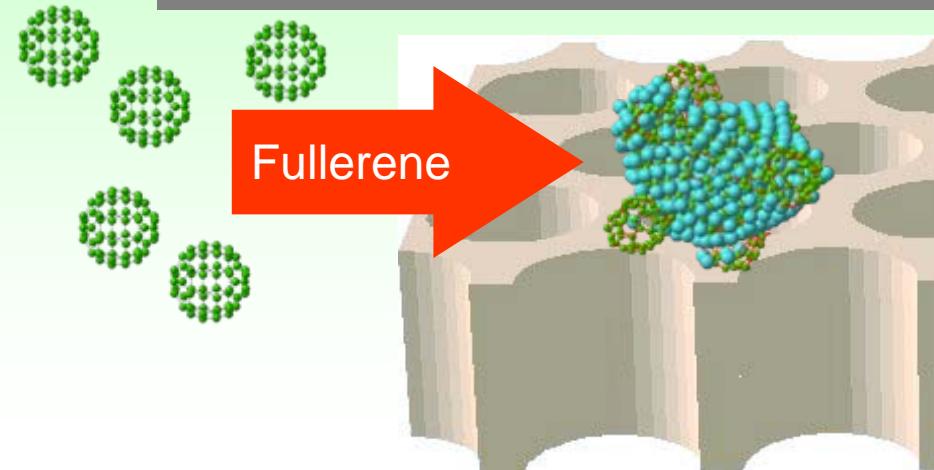
$(\text{CH}_3\text{CO}_2)_2\text{Fe}$
 $(\text{CH}_3\text{CO}_2)_2\text{Co} \cdot 4\text{H}_2\text{O}$ etc.

Supports

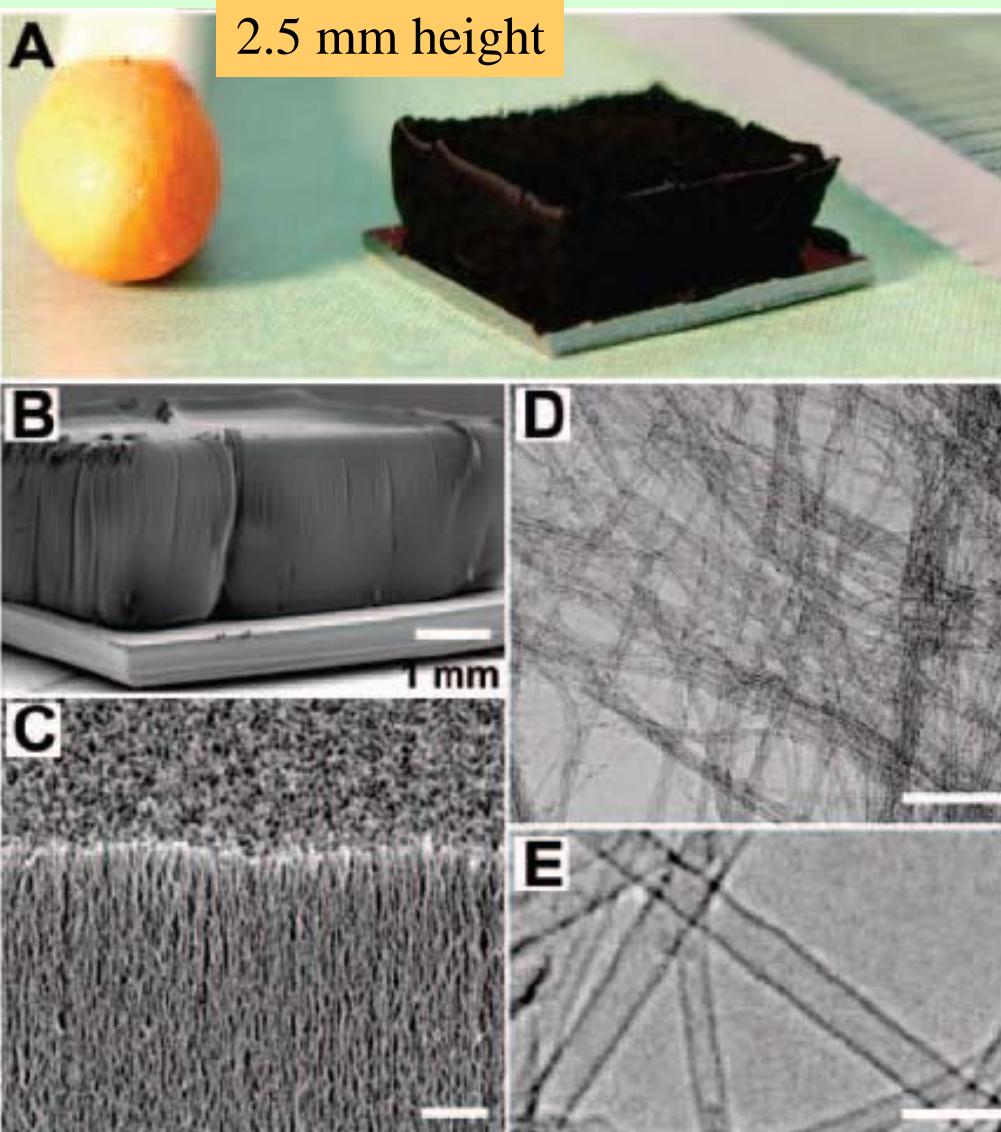
Zeolite USY
HSZ-390HUA



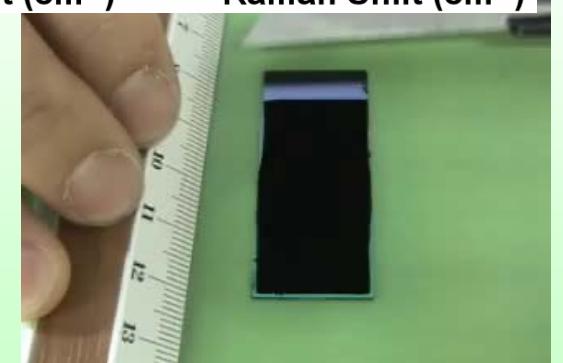
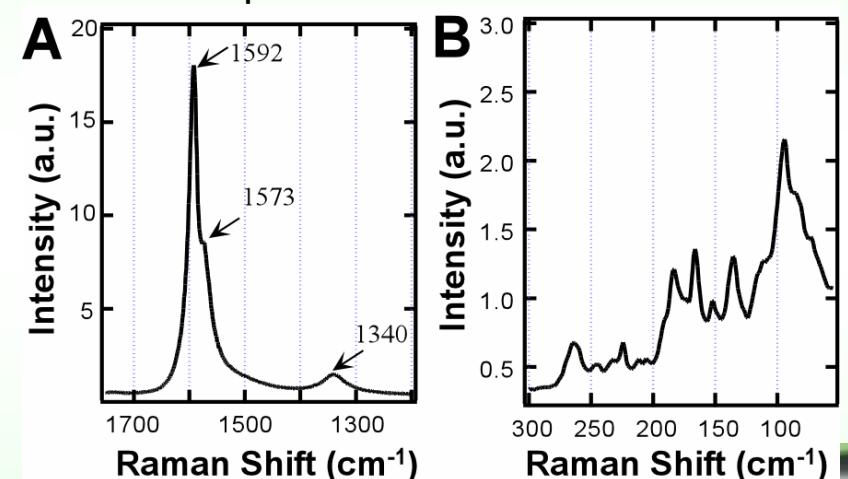
CCVD Generation of SWNTs from Fullerene



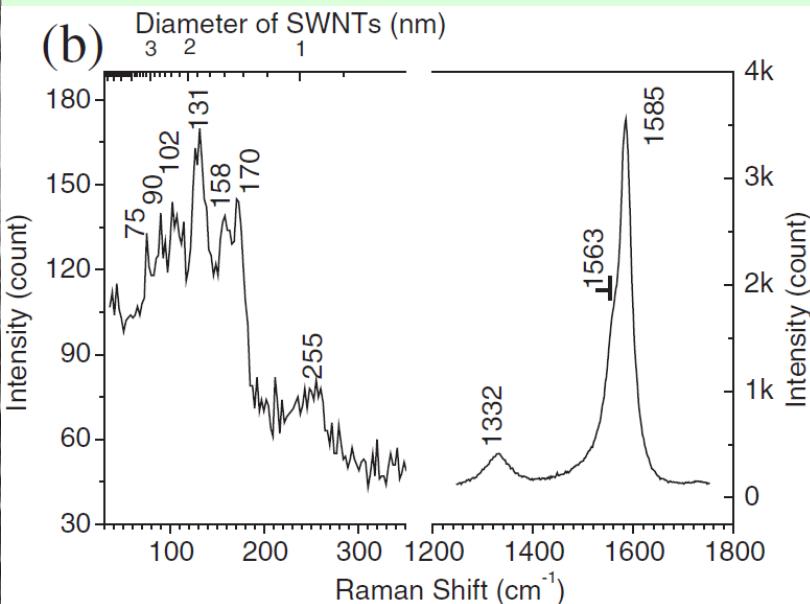
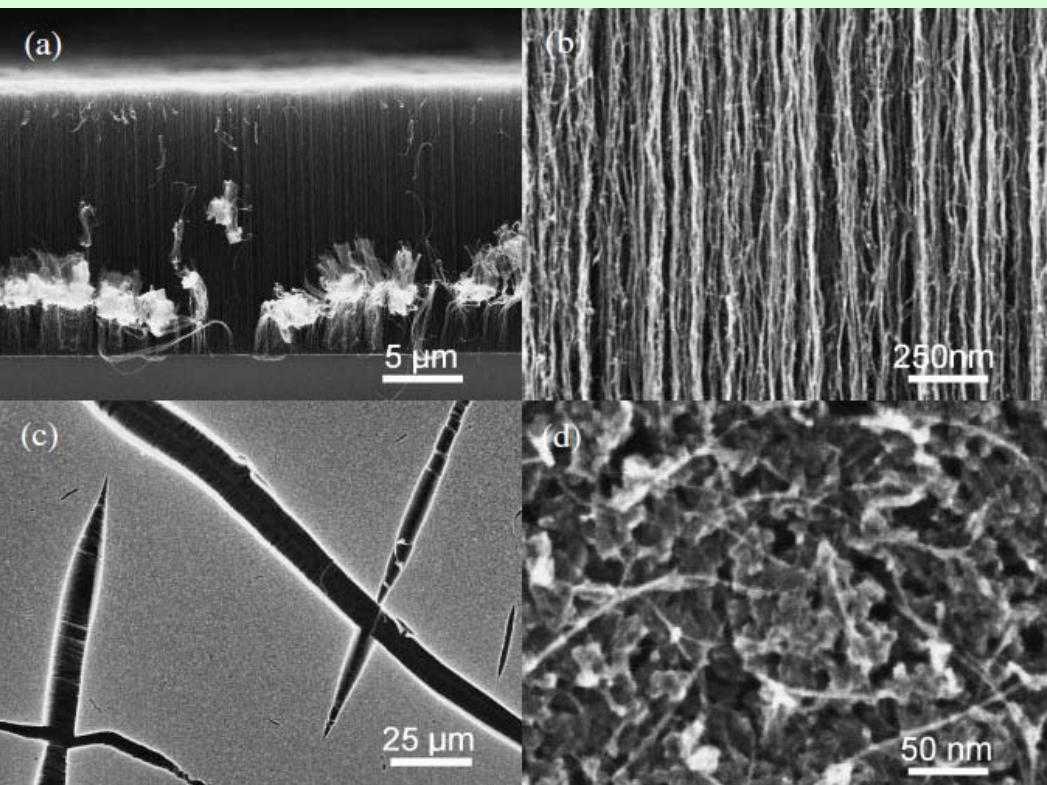
Water Assisted Super-Growth



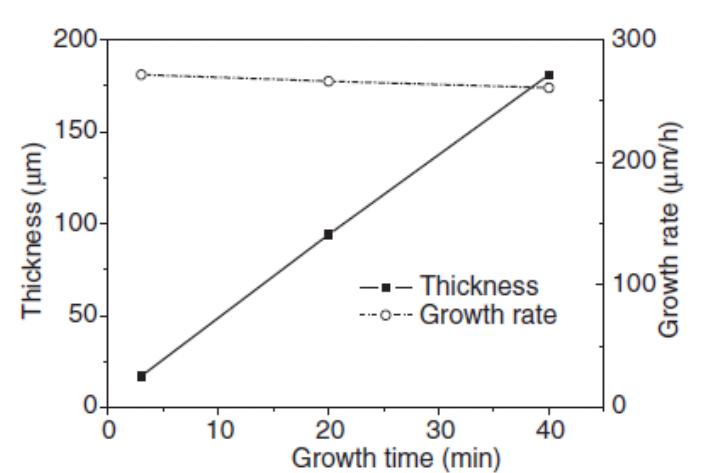
Catalysts: Al₂O₃ (10 nm)/Fe(1 nm)
Buffer: Ar (99.9999%) or He (99.9999%)
with 40% H₂ (99.99999%)
Water: 175 ppm
Source: 100 sccm ethylene (99.999%)
CVD Temp.: 750°C x 10 min



Super Growth by Kawarada Group



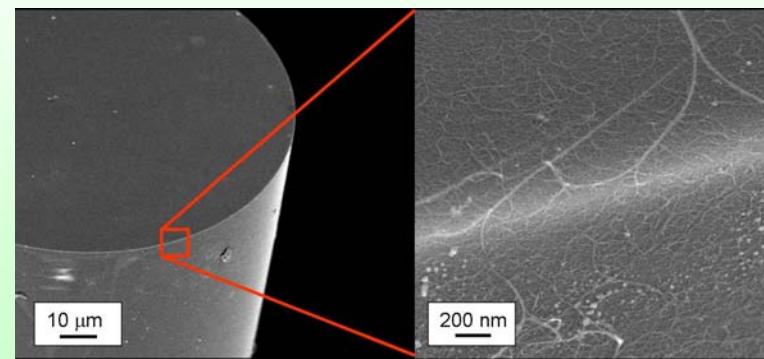
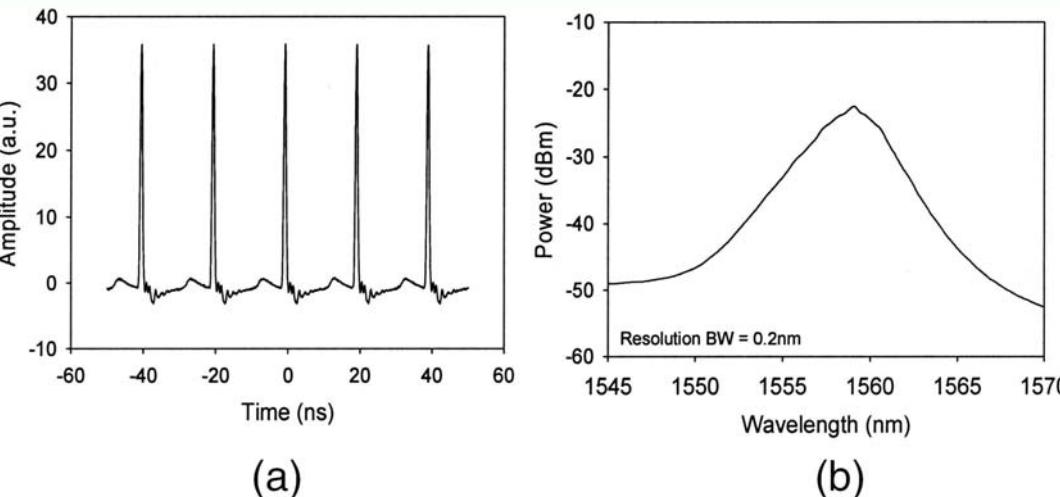
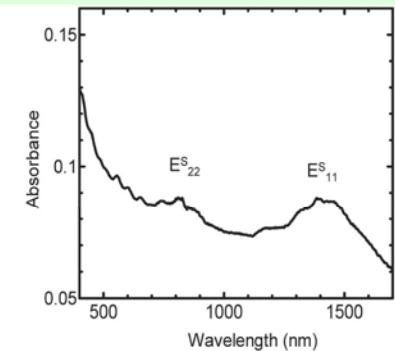
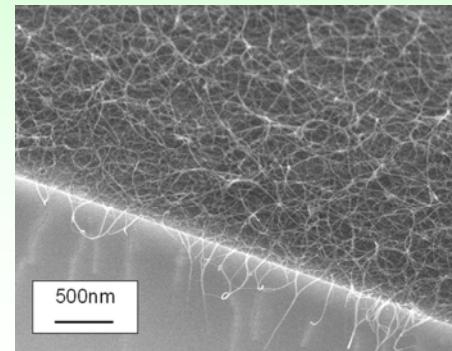
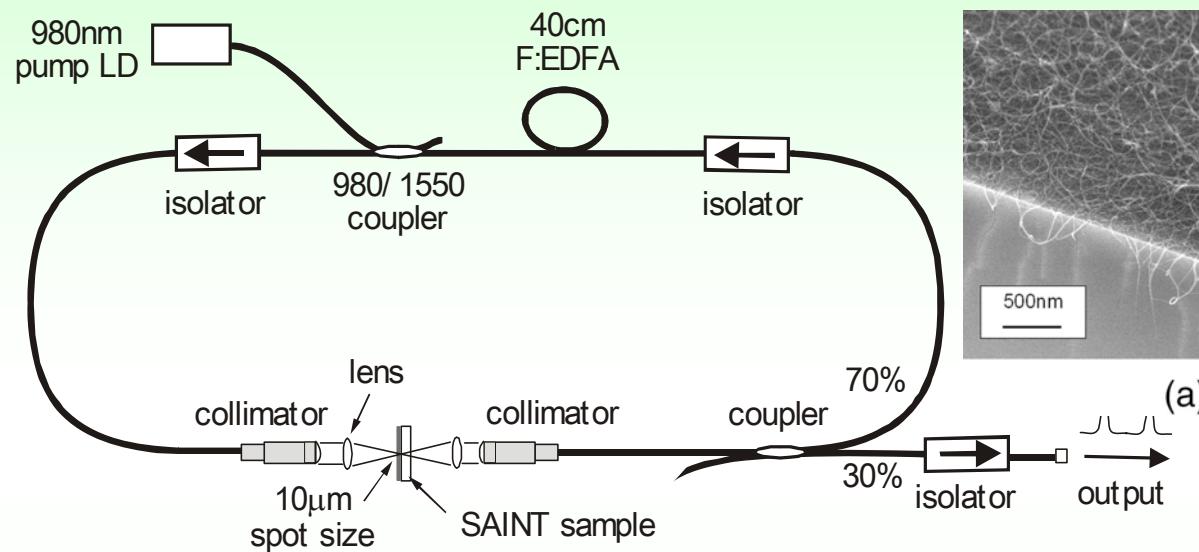
270 micron/h



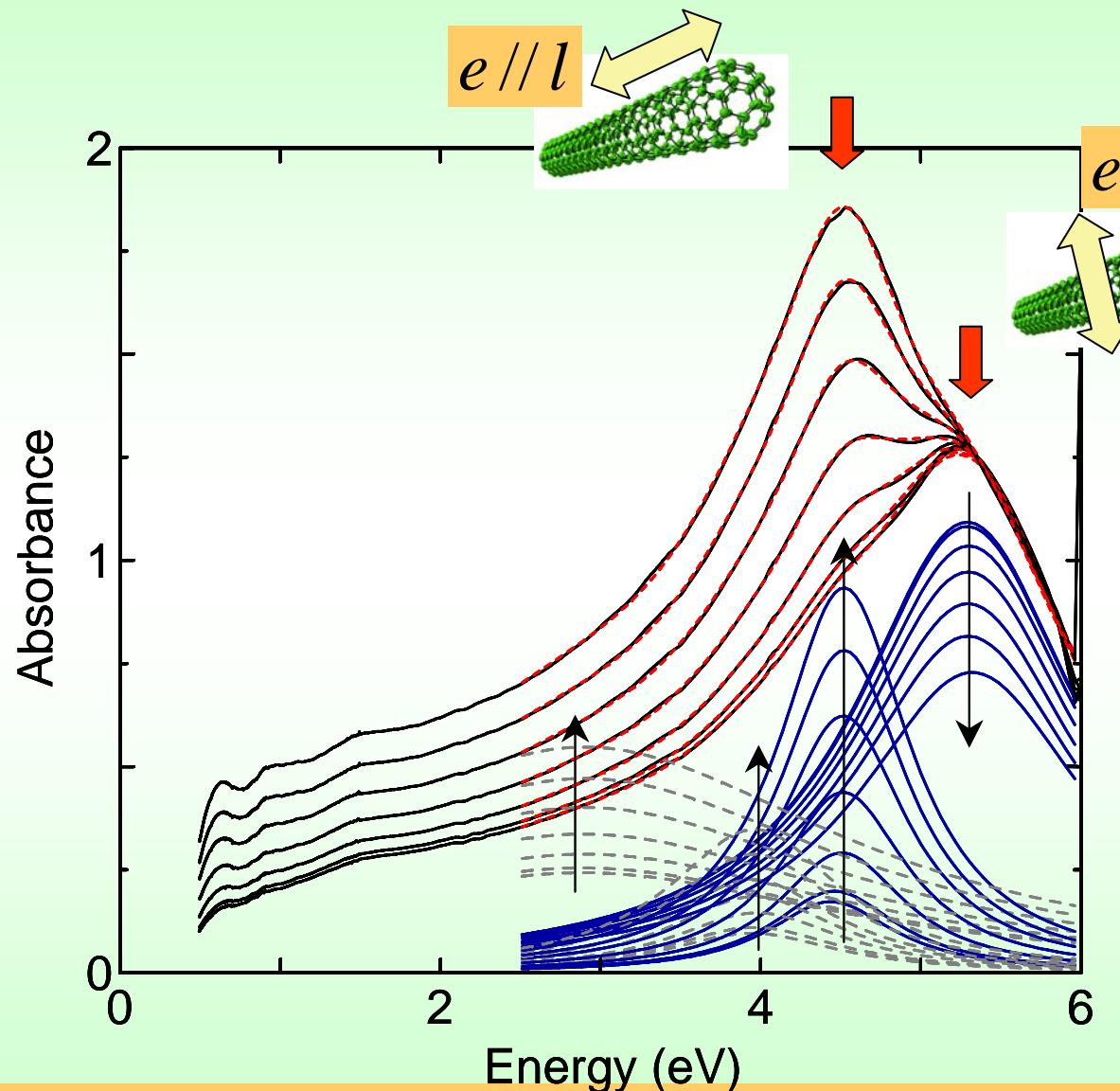
Point arc Microwave Plasma CVD
0.7nmAl₂O₃/0.5nmFe/5-70nmAl₂O₃ Catalysts

66kg/m³

Saturable Absorbers: Application to Mode-Locked Fiber Lasers

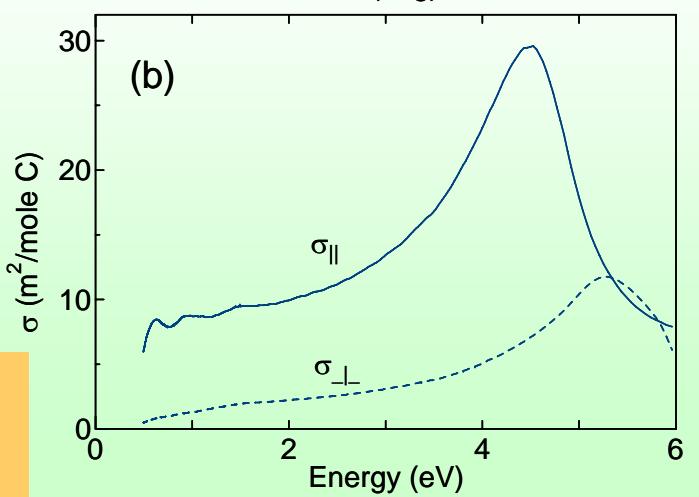
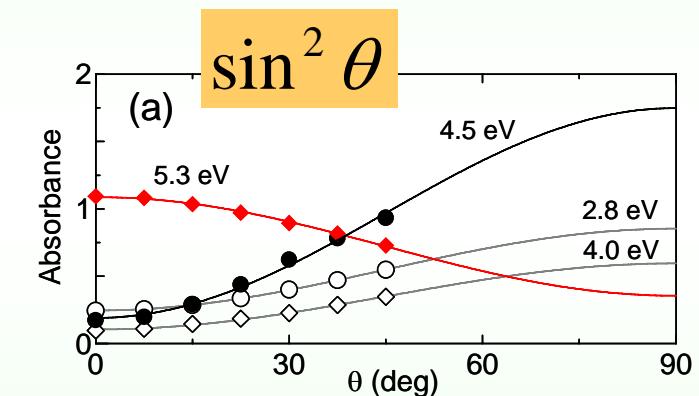


Polarized Optical Absorption of VA-SWNTs (2)

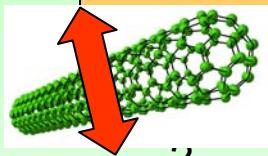


$$\sigma_{\parallel} = \frac{1}{3} \left\{ 2\eta_{\perp} + \eta_{\parallel} + \frac{2(\eta_{\parallel} - \eta_{\perp})}{S} \right\}$$

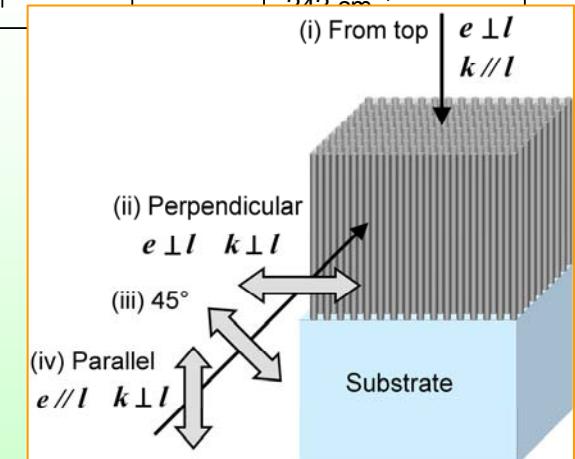
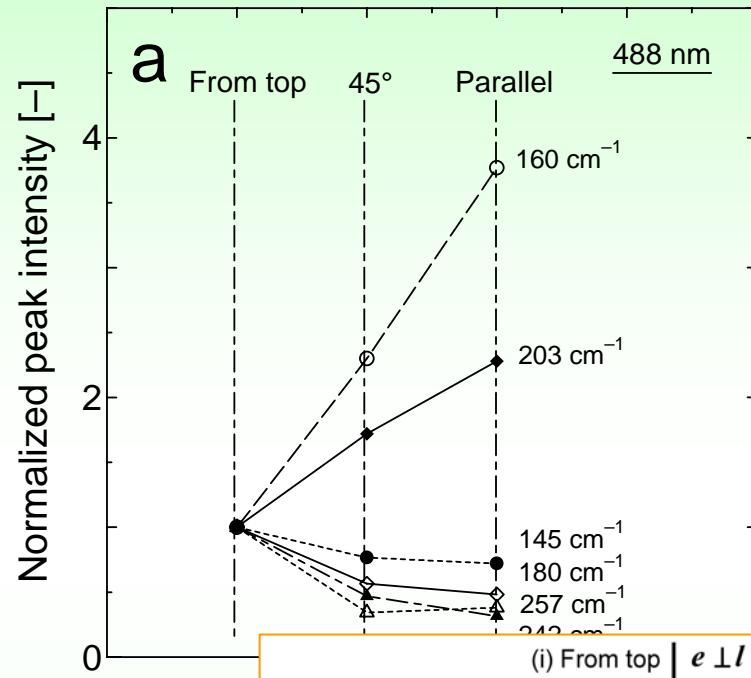
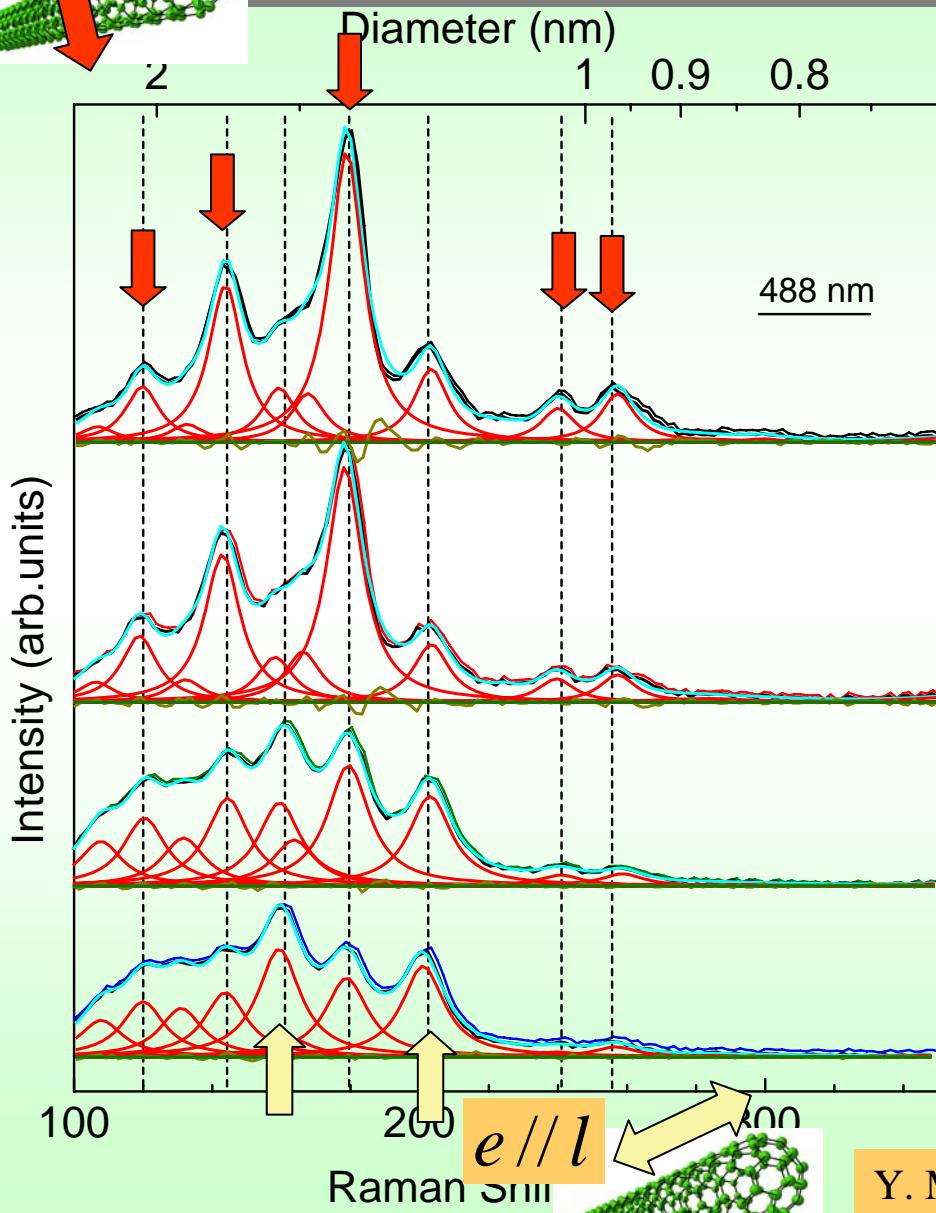
$$\sigma_{\perp} = \frac{1}{3} \left(2\eta_{\perp} + \eta_{\parallel} - \frac{\eta_{\parallel} - \eta_{\perp}}{S} \right)$$



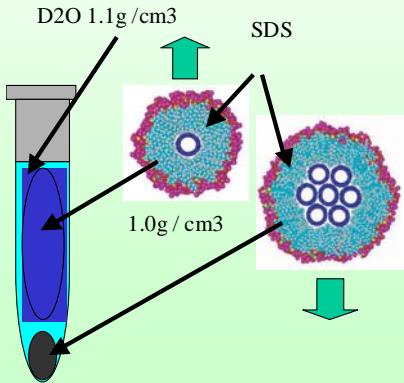
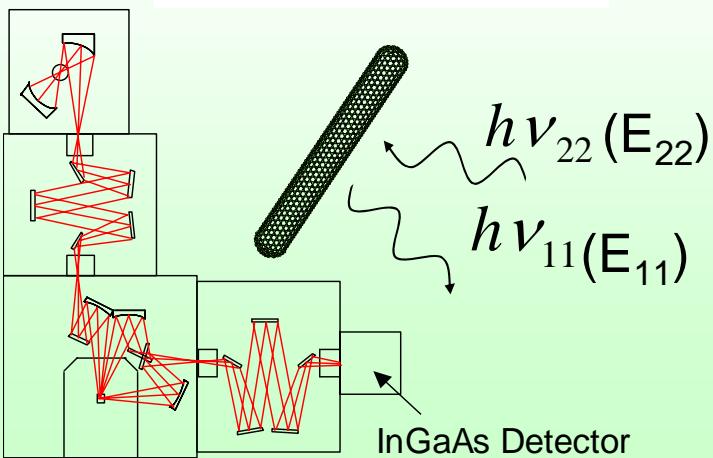
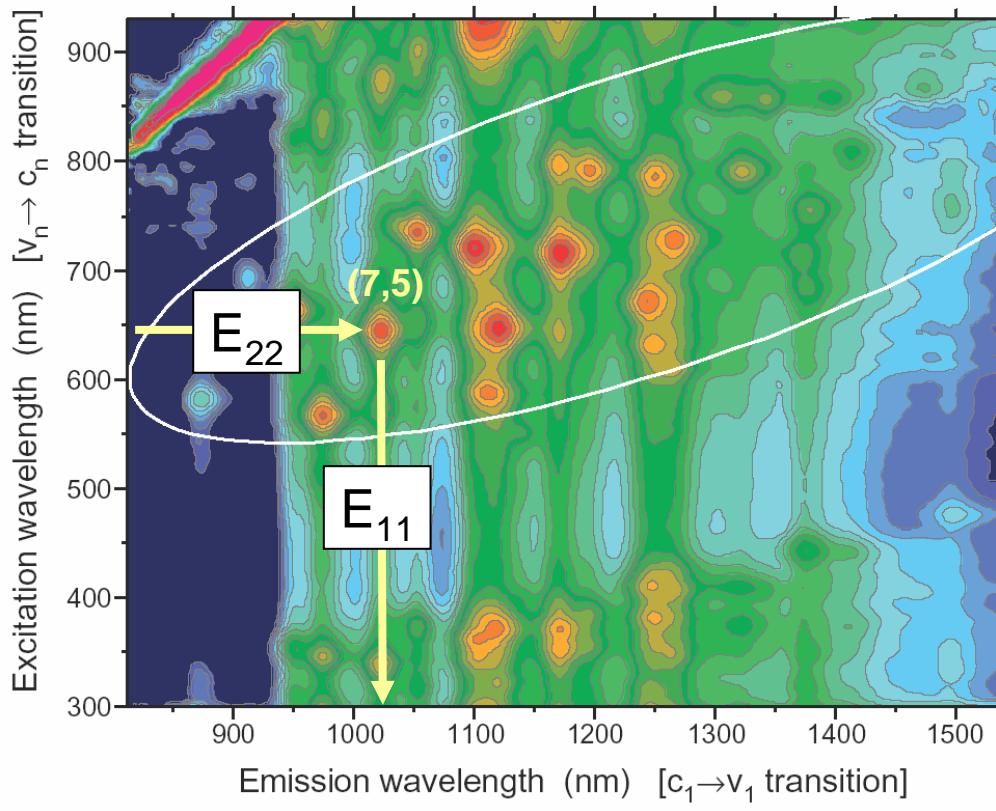
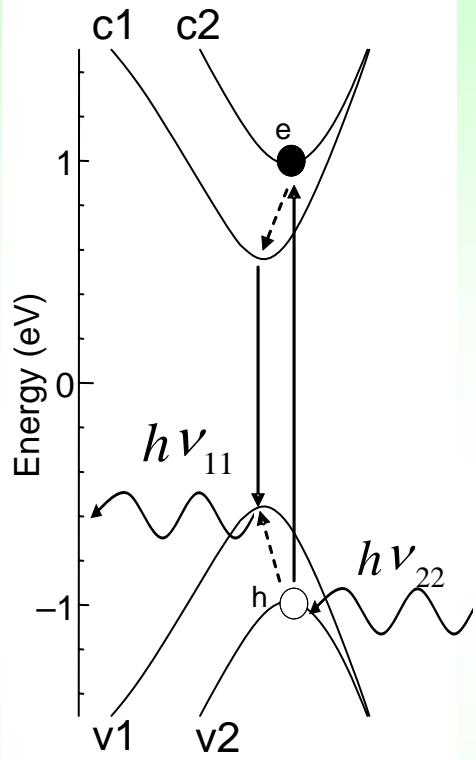
Y. Murakami, E. Einarsson, T. Edamura, S. Maruyama:
Phys. Rev. Lett., 94, 087402(2005)

$e \perp l$ 

Anomalous Raman Scattering



Photoluminescence Spectroscopy of SWNTs



M. J. O'Connell *et al.*, Science 297 (2002) 593

S. M. Bachilo *et al.*, Science 298 (2002) 2361

Strong sonication
Super centrifuge 380,000g x 1h